

# Photoemission Semiconductors

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## Band structure in 1-D

Consider an electron moving in a periodic potential  $V(x)$ . The period of the potential is  $a$ ,  $V(x + a) = V(x)$ . The Schrödinger equation for this case is,

$$-\frac{\hbar^2}{2m} \frac{d^2\psi}{dx^2} + V(x)\psi = E\psi.$$

Quantum mechanically, the electron moves as a wave through the potential. Due to the diffraction of these waves, there are bands of energies where the electron is allowed to propagate through the potential and bands of energies where no propagating solutions are possible. The Bloch theorem states that the propagating states have the form,

$$\psi = e^{ikx} u_k(x).$$

where  $k$  is the wavenumber and  $u_k(x)$  is a periodic function with periodicity  $a$ .

The solutions to the Schrödinger equation for a 1-D periodic potential can be calculated numerically. The following form can be used to calculate the dispersion relation between  $E$  and  $k$  for any one dimensional potential. Input the periodic potential  $V(x)$  in the interval between 0 and  $a$ .

The density of states is,

$$D(E) = \frac{2}{\pi} \frac{dk}{dE},$$

and the group velocity is,

$$v_g = \frac{1}{\hbar} \frac{dE}{dk}.$$

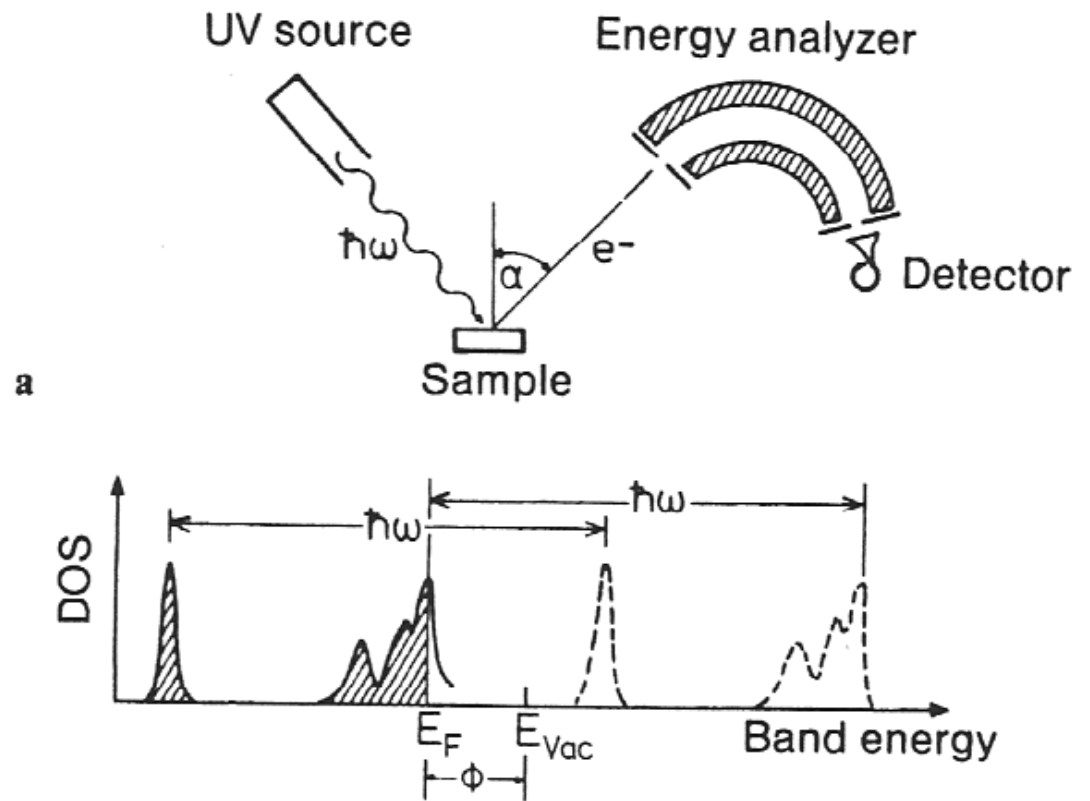
<http://lampx.tugraz.at/~hadley/ss1/bloch/bloch.php>

# Photoemission spectroscopy

UPS - Ultraviolet photoemission spectroscopy

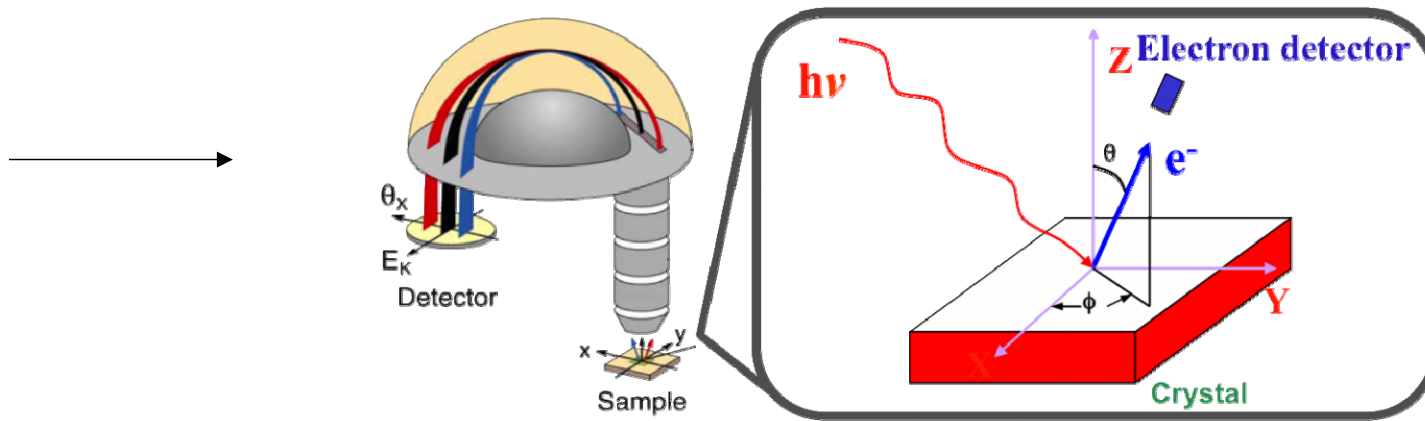
XPS - X-ray photoemission spectroscopy

Measure the density of states with photoemission spectroscopy

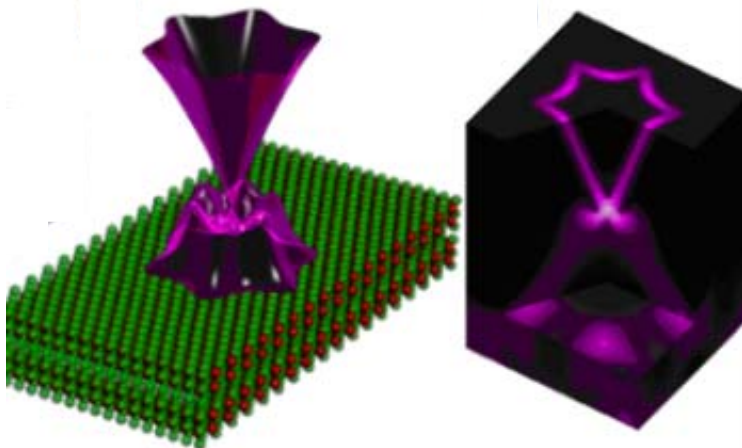


From: Ibach & Lueth

# Angle resolved photoemission spectroscopy (ARPES)

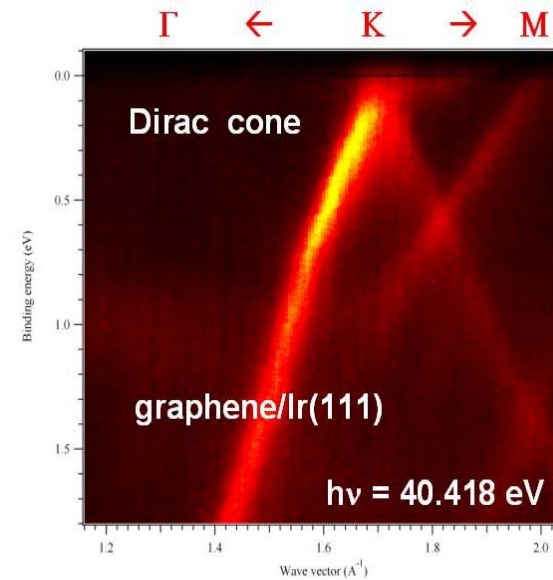


$\text{Bi}_2\text{Te}_3$

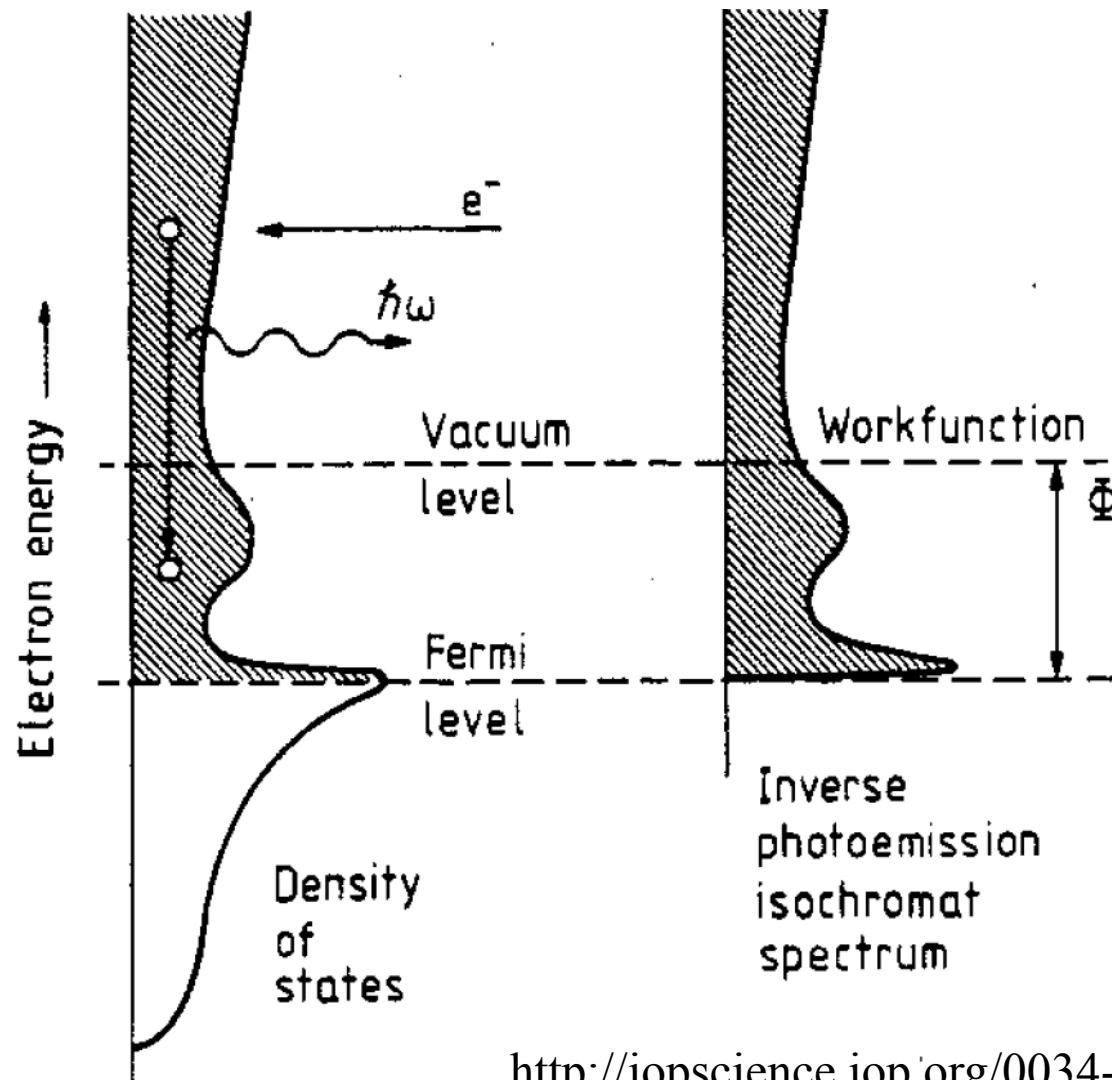


Topological insulator

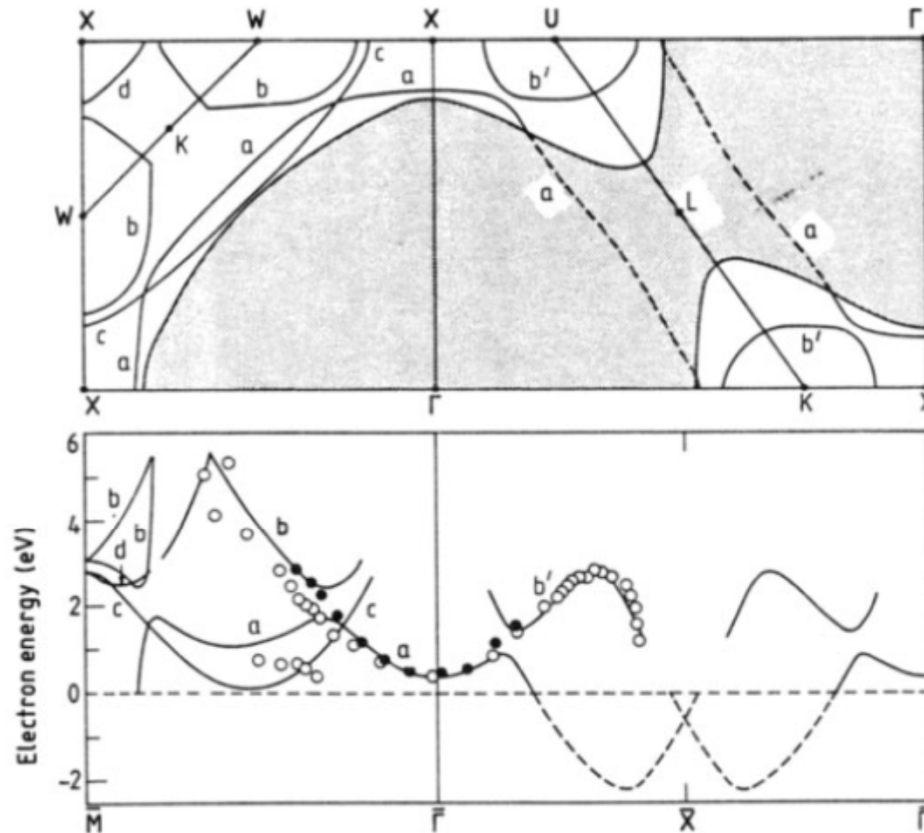
Measure the dispersion relation with angle resolved photoemission



# Inverse photoemission spectroscopy (IPES)



# $k$ -resolved Inverse Photoemission Spectroscopy (KRIPES)

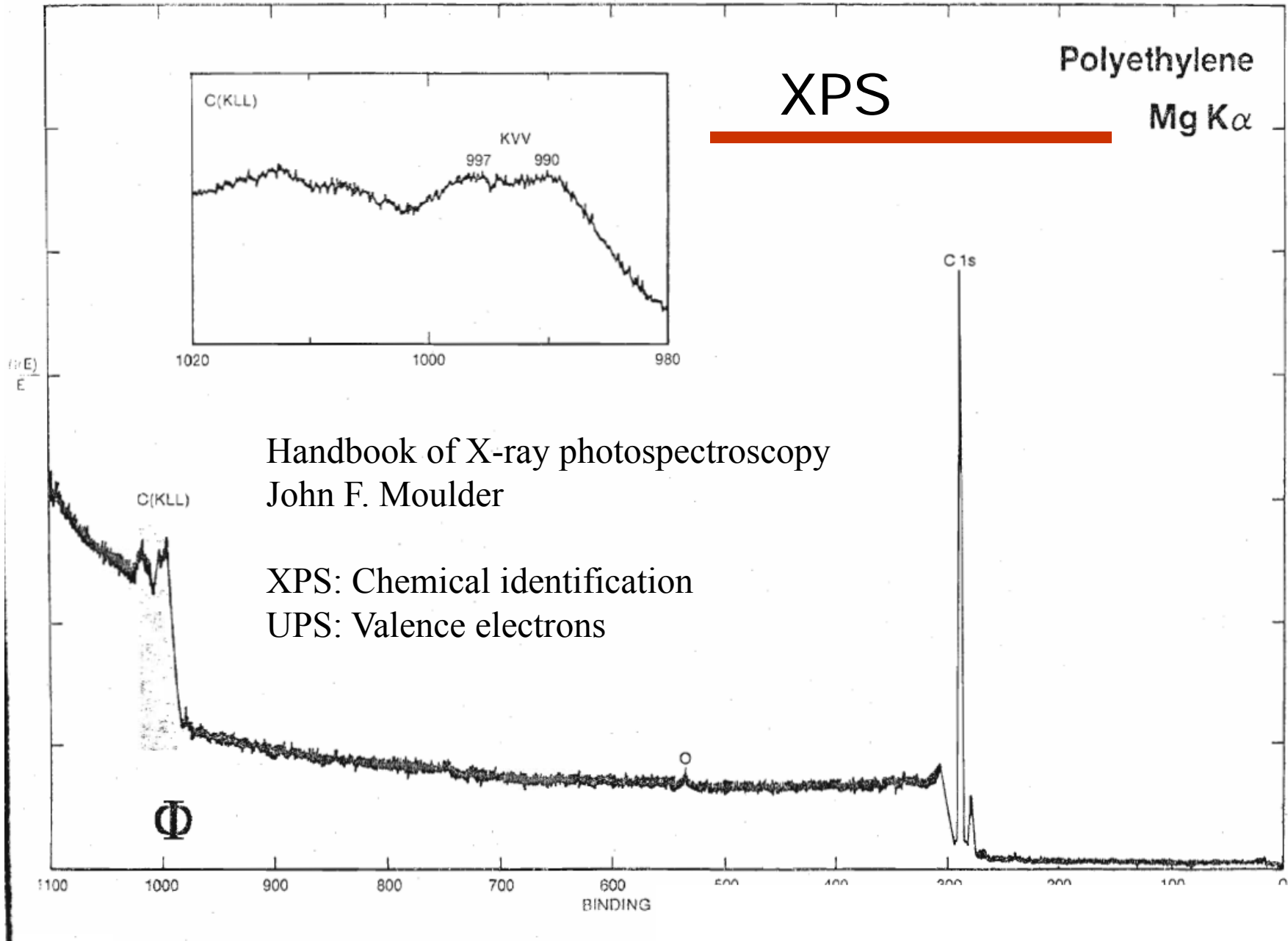


**Figure 9.** Band calculations and data for bulk direct transitions in the two principal azimuths  $\Gamma\bar{M}$  and  $\Gamma\bar{X}$  and Cu(001). Upper panel shows the Fermi surface and isochromat curves at  $\hbar\omega = 9.7$  eV for transitions into band 6. Lower panel shows the corresponding  $E_t(k_{||})$  projections. Computations and filled data circles are from Woodruff *et al* (1982); open circles are data from Jacob *et al* (1986).

Polyethylene

XPS

Mg K $\alpha$



Handbook of X-ray photospectroscopy  
John F. Moulder

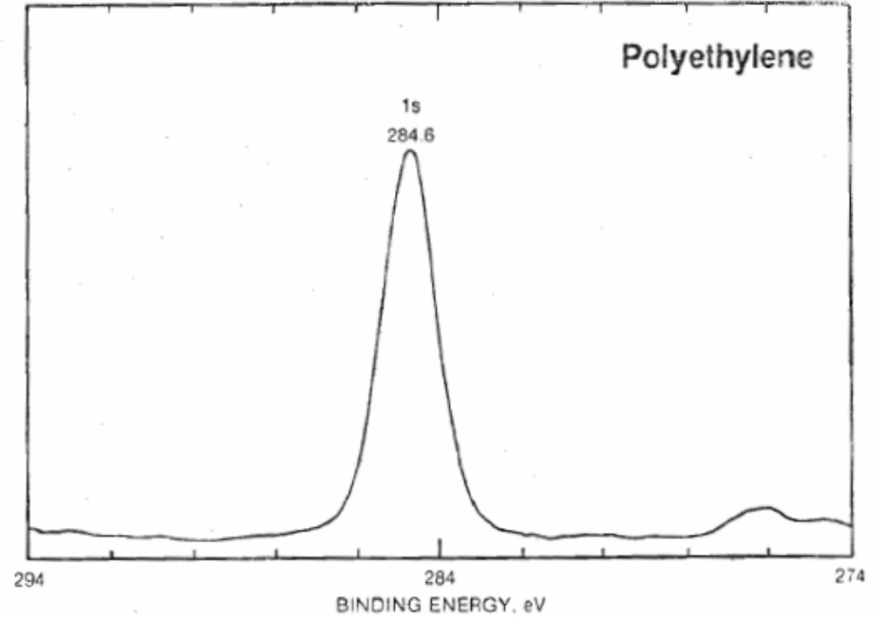
XPS: Chemical identification  
UPS: Valence electrons

# XPS

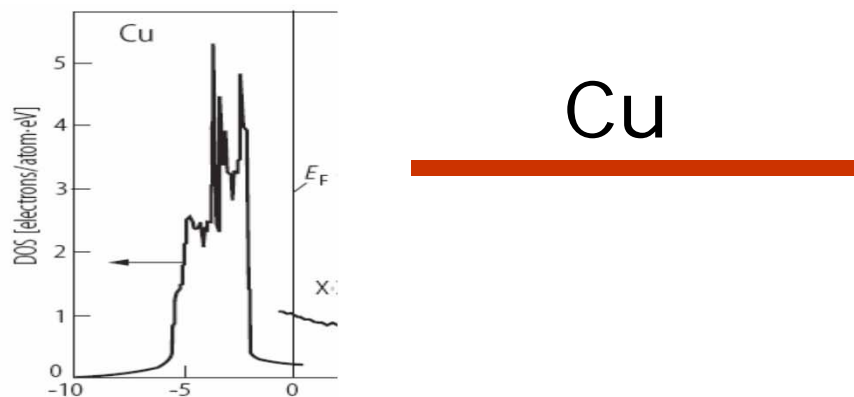
## Carbon, C Atomic Number 6

HANDBOOK OF X-RAY PHOTOELECTRON SPECTROSCOPY

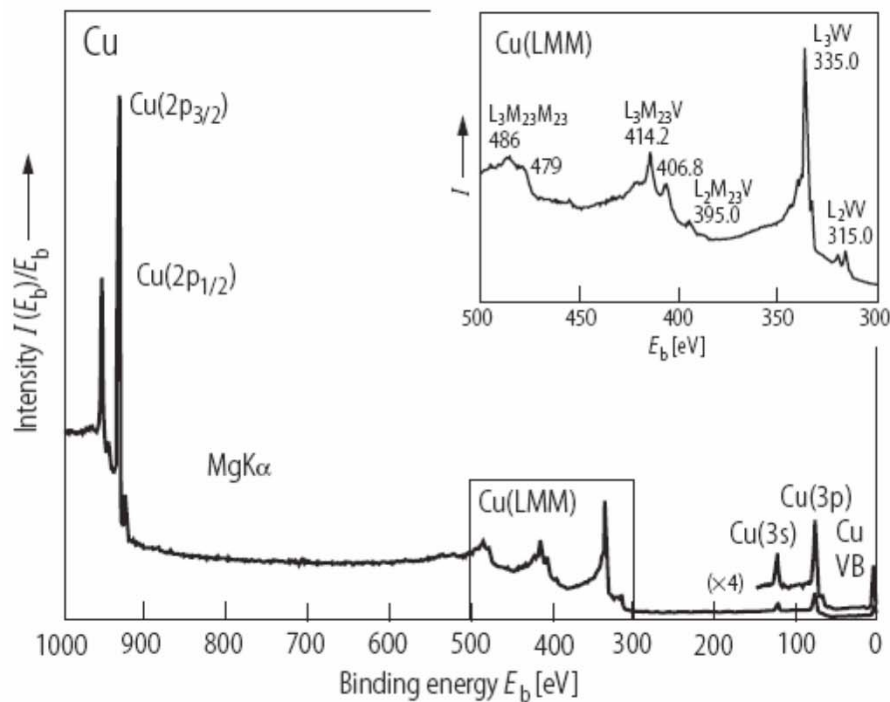
COMPOUND	1s BINDING ENERGY, eV					REF.
	280	284	288	292	296	
HfC						RH1
TiC						RH1
WC						RH1
C (graphite)						HJG
(CH <sub>2</sub> ) <sub>n</sub>						Φ
Mn(C <sub>5</sub> H <sub>5</sub> ) <sub>2</sub>						BCD
SnPh <sub>4</sub>						BAL
MeCH <sub>2</sub> NH <sub>2</sub>						GHH
Cr(C <sub>6</sub> H <sub>6</sub> ) <sub>2</sub>						PFD
MeCH <sub>2</sub> Cl						GHH
MeCH <sub>2</sub> OH						GHH
MeCH <sub>2</sub> OEt						GHH
MeCH <sub>2</sub> OOCMe						GHH
CS <sub>2</sub>						GHH
Fe(CO) <sub>5</sub>						BC1
Me <sub>2</sub> CO						GHH
(NH <sub>2</sub> ) <sub>2</sub> CO						GHH
C <sub>5</sub> F <sub>6</sub>						GHH
MeCOONa						GHH
MeCOOEt						GHH
MeCOOH						GHH
Na <sub>2</sub> CO <sub>3</sub>						GHH
NaHCO <sub>3</sub>						GHH
CO						BC1
CO <sub>2</sub>						GHH
(CHFCH <sub>2</sub> ) <sub>n</sub>						CFK
(CHFCHF) <sub>n</sub>						CFK
(CHFCH <sub>2</sub> ) <sub>n</sub>						CFK
(CF <sub>2</sub> CH <sub>2</sub> ) <sub>n</sub>						CFK
(CF <sub>2</sub> CHF) <sub>n</sub>						CFK
(CF <sub>2</sub> ) <sub>n</sub>						CFK
CF <sub>3</sub> COONa						GHH
CCl <sub>4</sub>						GHH
CF <sub>3</sub> COMe						GHH
CF <sub>3</sub> COOEt						GHH



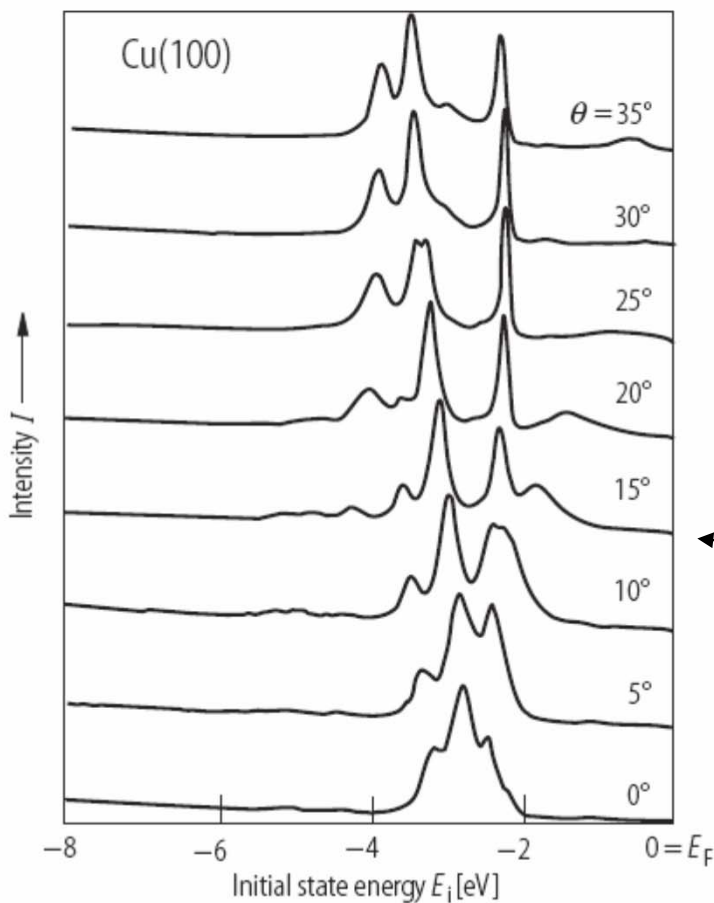




Cu



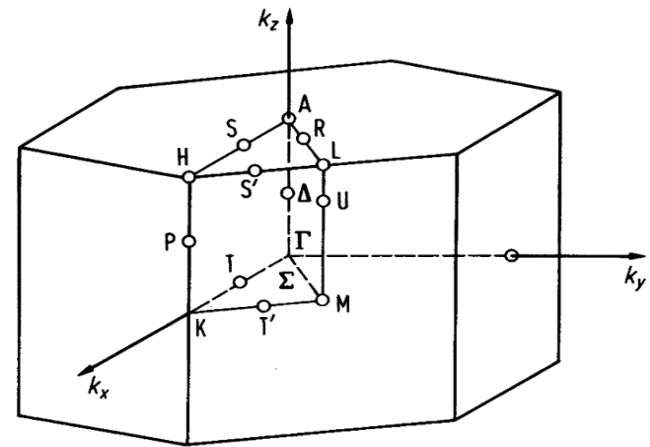
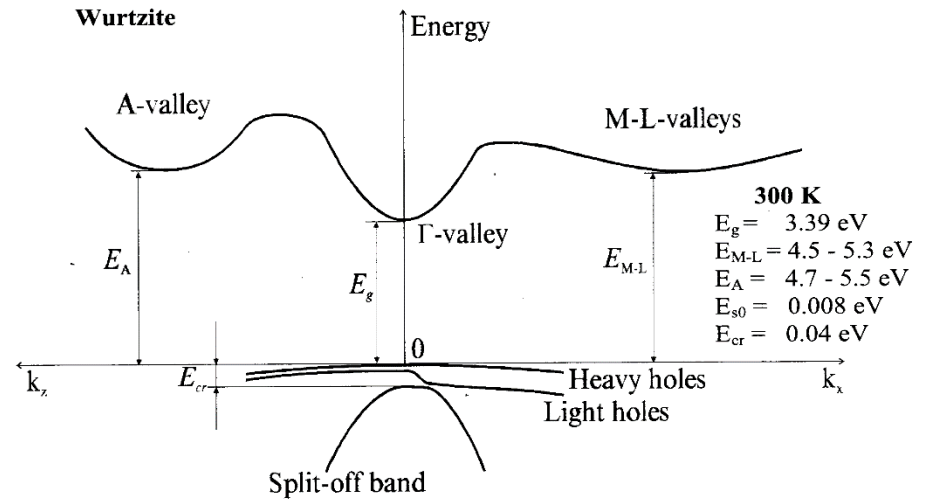
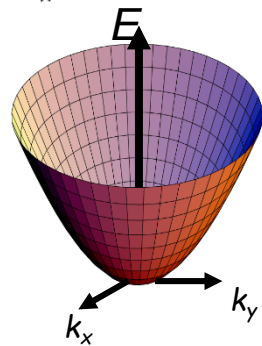
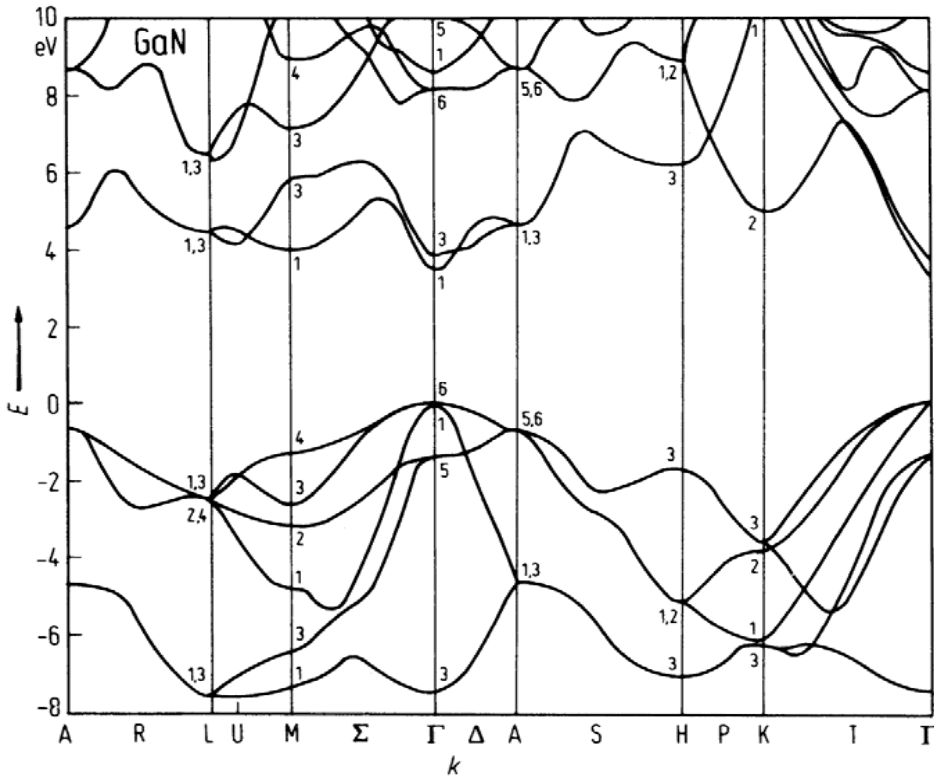
XPS



ARPES

**Fig. 28.** Cu(100). Angle-resolved photoelectron spectra taken at different polar angles  $\theta$  along the  $\Gamma XUL$  bulk mirror plane. Photon energy  $h\nu = 21.2$  eV, sample temperature  $T = 50$  K [93M1]. For further data taken at room temperature see [79H1]. For data taken with linear-polarized photons at  $h\nu = 40^\circ$  see [83G].

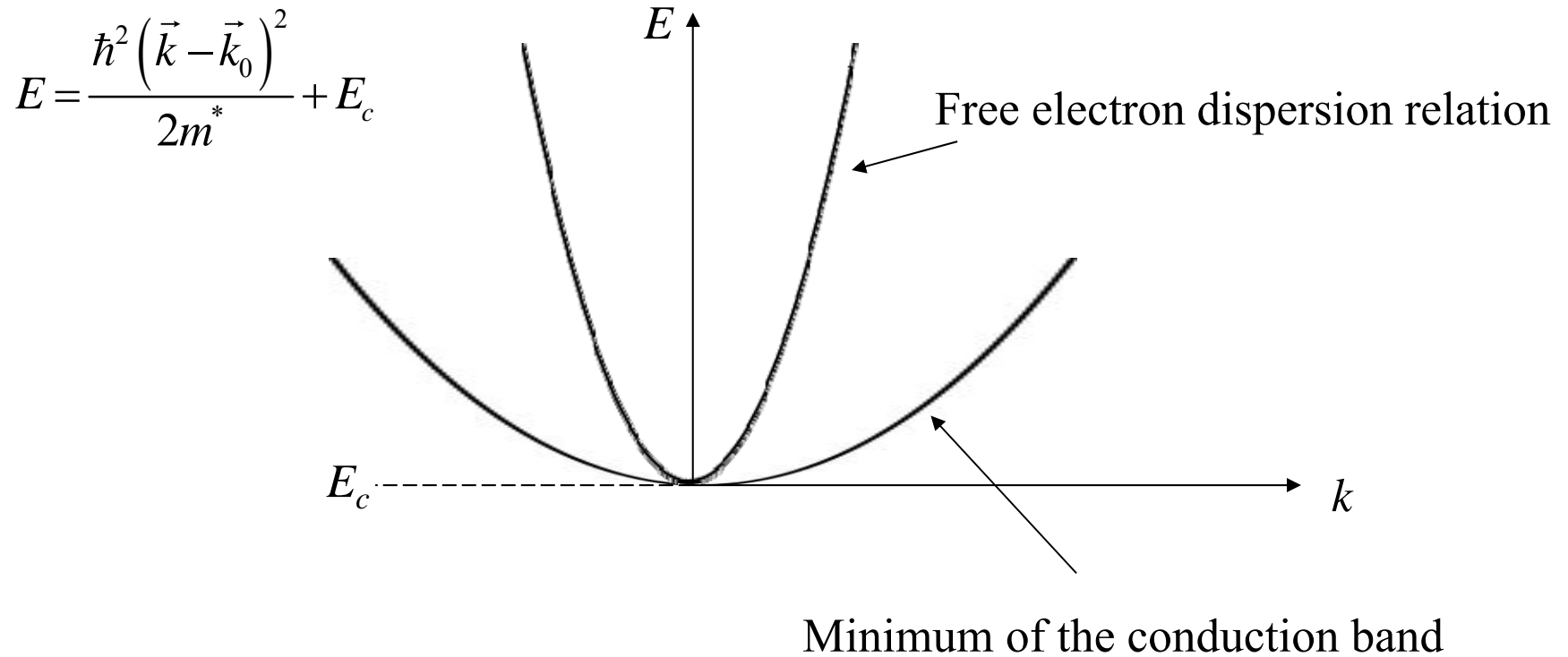
# GaN



1st Brillouin zone of hcp

# Conduction band minimum

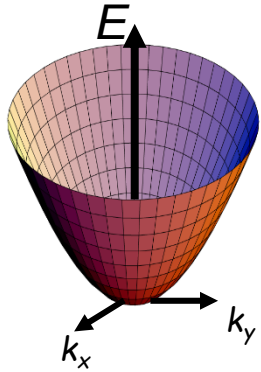
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Near the conduction band minimum, the bands are approximately parabolic.

# Effective mass

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$$E = \frac{\hbar^2 (\vec{k} - \vec{k}_0)^2}{2m^*} + E_c$$

The parabola at the bottom of the conduction band does not have the same curvature as the free-electron dispersion relation. We define an effective mass to characterize the conduction band minimum.

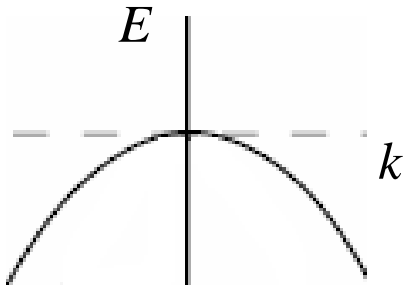
$$m^* = \frac{\hbar^2}{\frac{d^2 E(\vec{k})}{dk_x^2}}$$

This effective mass is used to describe the response of electrons to external forces in the particle picture.

# Top of the valence band

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In the valence band, the effective mass is negative.



$$m^* = \frac{\hbar^2}{\frac{d^2 E(\vec{k})}{dk_x^2}} < 0$$

Charge carriers in the valence band are positively charged holes.

$m_h^*$  = effective mass of holes

$$m_h^* = \frac{-\hbar^2}{\frac{d^2 E(\vec{k})}{dk_x^2}}$$

# Holes

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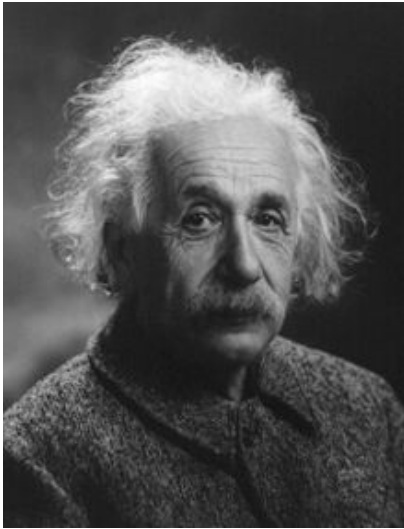
A completely filled band does not contribute to the current.

$$\begin{aligned}\vec{j} &= \int_{\text{filled states}} -e\vec{v}(\vec{k})D(\vec{k})f(\vec{k})d\vec{k} \\ &= \int_{\text{band}} -e\vec{v}(\vec{k})D(\vec{k})f(\vec{k})d\vec{k} - \int_{\text{empty states}} -e\vec{v}(\vec{k})D(\vec{k})f(\vec{k})d\vec{k} \\ &= \int_{\text{empty states}} e\vec{v}(\vec{k})D(\vec{k})f(\vec{k})d\vec{k}\end{aligned}$$

Holes have a positive charge and a positive mass.

# Holes

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Albert Einstein



Erwin Schrödinger



Paul Adrien Maurice Dirac