

# 18. Electrons

---

May 29, 2018

# Fermi energy

---

In solid state physics books,

$$E_F = \mu(T=0).$$

In semiconductor books,  $E_F(T) = \mu(T)$ .

$$\text{At } T = 0 \quad n = \int_{-\infty}^{E_F} D(E) dE$$

In one dimension,

$$n = \int_0^{E_F} \frac{1}{\pi\hbar} \sqrt{\frac{2m}{E}} dE = \frac{2}{\pi\hbar} \sqrt{2mE_F}$$

$$E_F = \frac{\pi^2 \hbar^2 n^2}{8m}$$

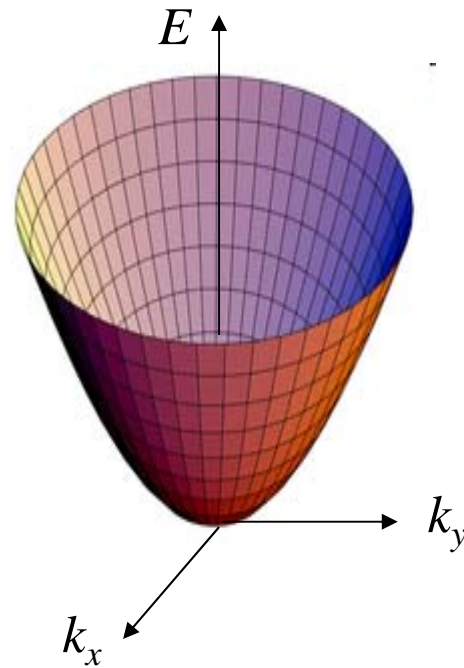
# Free particles in 2-d

---

Density of states

$$E = \frac{\hbar^2 k^2}{2m}$$

$$k = \frac{\sqrt{2mE}}{\hbar}$$



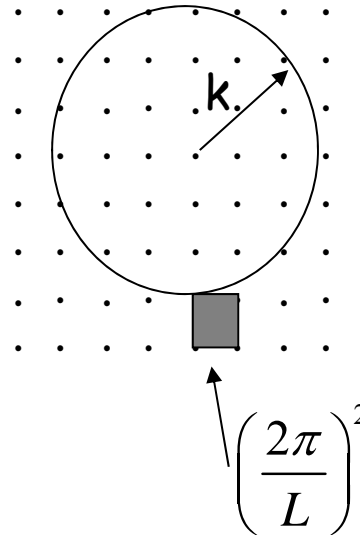
$$\frac{dk}{dE} = \frac{1}{2\hbar} \sqrt{\frac{2m}{E}}$$

# Free particles in 2-d

$$E = \frac{\hbar^2 k^2}{2m}$$

$$k = \frac{\sqrt{2mE}}{\hbar}$$

$$\frac{dk}{dE} = \frac{1}{2\hbar} \sqrt{\frac{2m}{E}}$$



$$D(E) = D(k) \frac{dk}{dE}$$

$$L^2 D(k) dk = \frac{2 \cdot 2\pi k dk}{\left(\frac{2\pi}{L}\right)^2}$$

$$D(k) = \frac{k}{\pi} \text{ m}^{-1}$$

$$D(E) = \frac{k}{\pi} \frac{dk}{dE} = \frac{\sqrt{2mE}}{\hbar\pi} \frac{1}{2\hbar} \sqrt{\frac{2m}{E}}$$

$$D(E) = \frac{m}{\pi\hbar^2}$$

# Free particles in 2-d

---

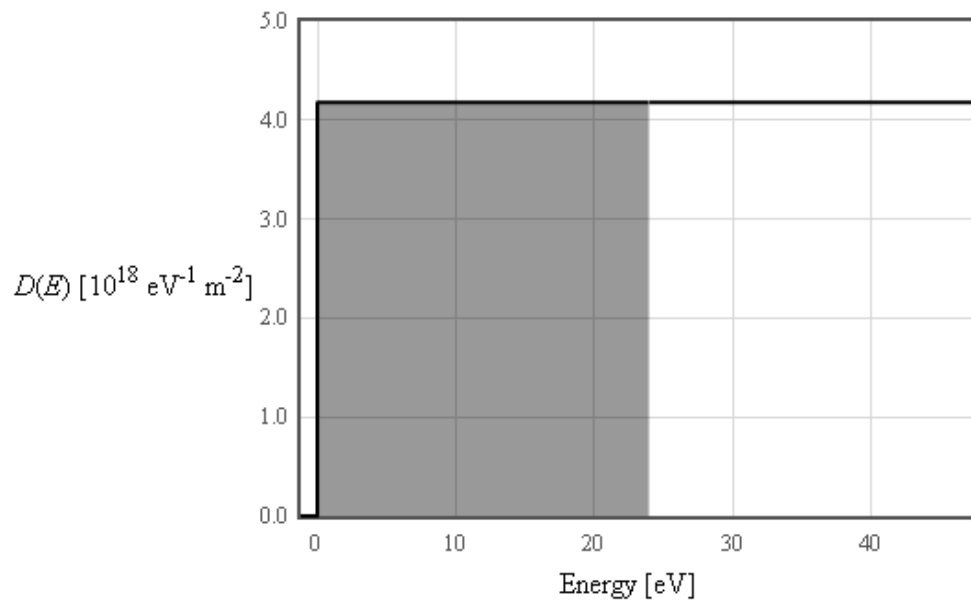
$$D(E) = \frac{m}{\pi \hbar^2}$$

At  $T = 0$ :

$$n = \int_0^{E_F} D(E) dE$$

$$n = \frac{N}{L^2} = \frac{m}{\pi \hbar^2} \int_0^{E_F} dE = \frac{m}{\pi \hbar^2} E_F$$

$$E_F = \frac{\pi \hbar^2 n}{m}$$



# Fermi circle

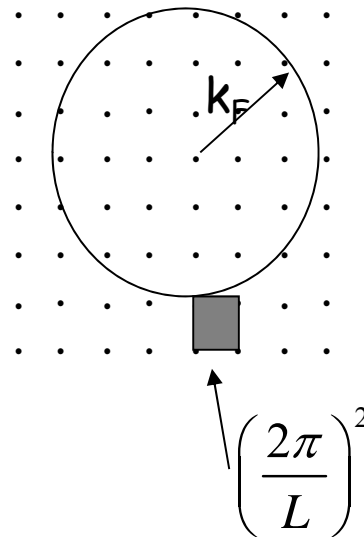
---

$$N = \frac{2\pi k_F^2}{\left(\frac{2\pi}{L}\right)^2}$$

$n = N/L^2 =$  electron density

$$k_F = \sqrt{2\pi n}$$

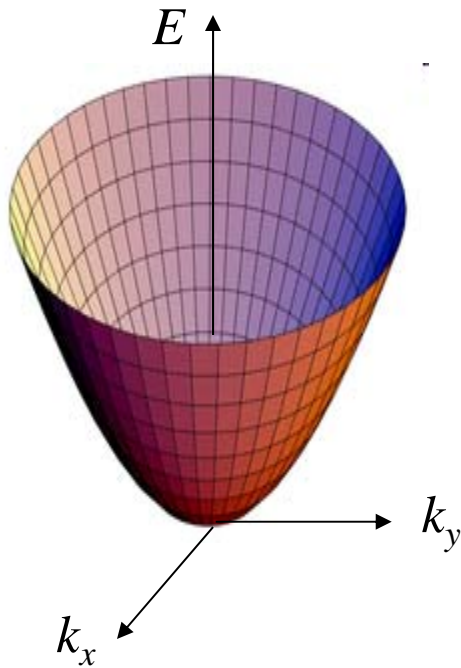
$$E_F = \frac{\hbar^2 k_F^2}{2m} = \frac{\pi \hbar^2 n}{m}$$



At  $T = 0$ , all states inside the Fermi circle are occupied and those outside are empty.

# Free particles in 3-d

$$E = \frac{\hbar^2 k^2}{2m}$$



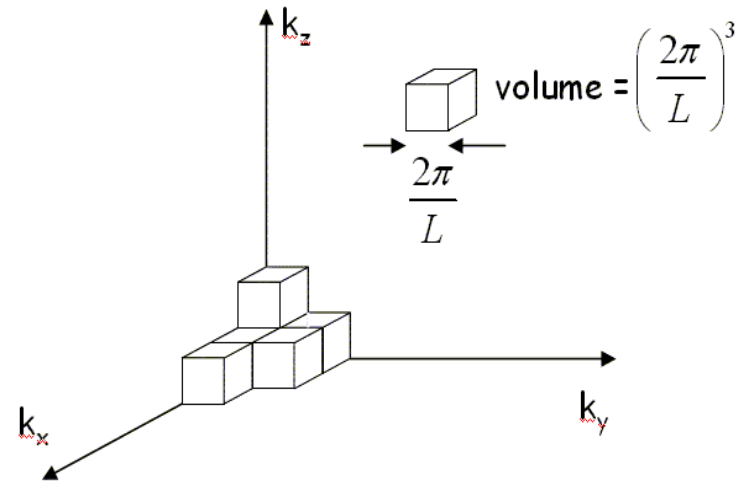
Density of states

$$D(k) = \frac{k^2}{\pi^2}$$

$$\frac{dk}{dE} = \frac{1}{2\hbar} \sqrt{\frac{2m}{E}}$$

$$k^2 = \frac{2mE}{\hbar^2}$$

$$D(E) = D(k) \frac{dk}{dE}$$

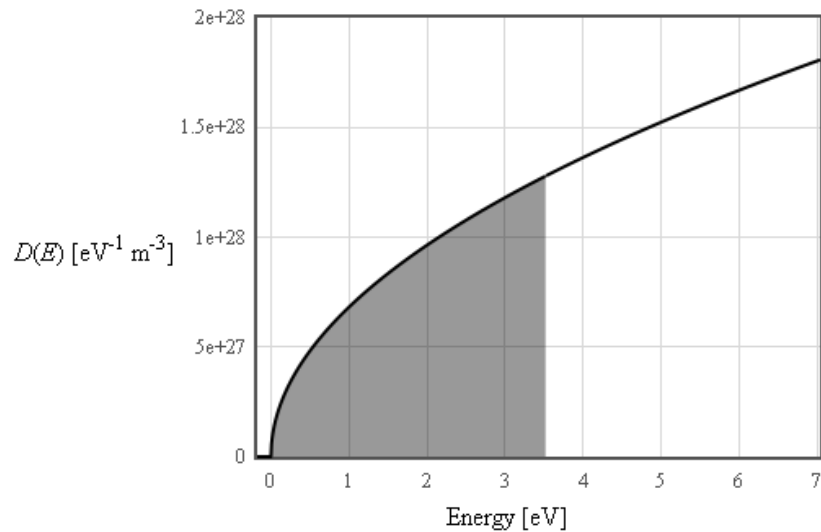


$$k_x, k_y, k_z = \dots, \frac{-4\pi}{L}, \frac{-2\pi}{L}, 0, \frac{2\pi}{L}, \frac{4\pi}{L}, \dots$$

$$D(E) = \frac{(2m)^{\frac{3}{2}}}{2\pi^2 \hbar^3} \sqrt{E}$$

# Free particles in 3-d

---



At  $T = 0$ :

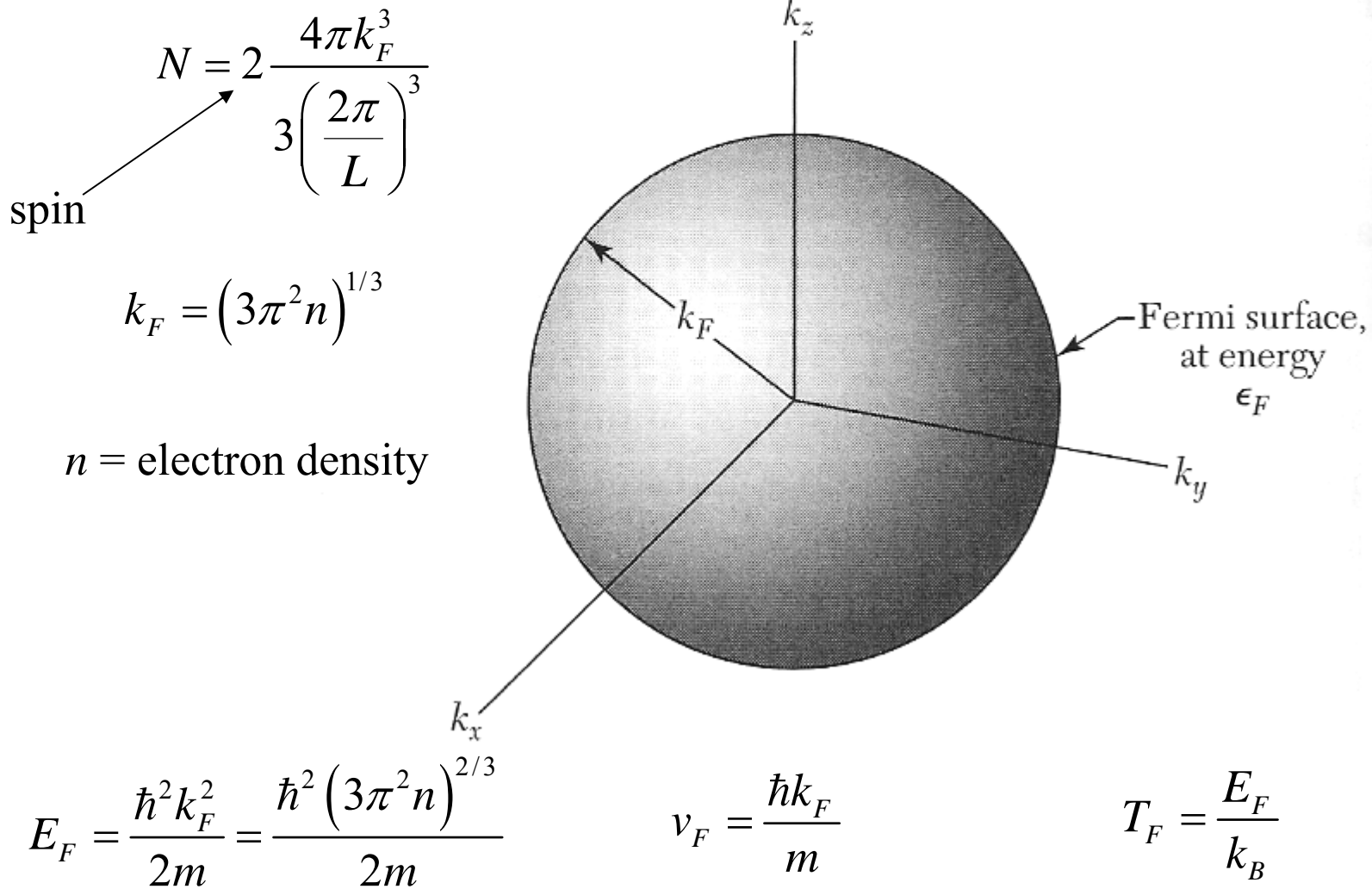
$$n = \int_0^{E_F} D(E) dE$$

$$n = \frac{N}{L^3} = \frac{\sqrt{2}m^{3/2}}{\pi^2\hbar^3} \int_0^{E_F} \sqrt{E} dE = \frac{(2m)^{3/2}}{3\pi^2\hbar^3} E_F^{3/2}$$

$$E_F = \frac{\hbar^2}{2m} (3\pi^2 n)^{2/3}$$



# Fermi sphere



The thermal and electronic properties depend on the states at the Fermi surface.

**Table 1** Calculated free electron Fermi surface parameters for metals at room temperature

(Except for Na, K, Rb, Cs at 5 K and Li at 78 K)

Valency	Metal	Electron concentration, in $\text{cm}^{-3}$	Radius <sup>a</sup> parameter $r_n$	Fermi wavevector, in $\text{cm}^{-1}$	Fermi velocity, in $\text{cm s}^{-1}$	Fermi energy, in eV	Fermi temperature $T_F \equiv \epsilon_F/k_B$ , in deg K
1	Li	$4.70 \times 10^{22}$	3.25	$1.11 \times 10^8$	$1.29 \times 10^8$	4.72	$5.48 \times 10^4$
	Na	2.65	3.93	0.92	1.07	3.23	3.75
	K	1.40	4.86	0.75	0.86	2.12	2.46
	Rb	1.15	5.20	0.70	0.81	1.85	2.15
	Cs	0.91	5.63	0.64	0.75	1.58	1.83
	Cu	8.45	2.67	1.36	1.57	7.00	8.12
	Ag	5.85	3.02	1.20	1.39	5.48	6.36
	Au	5.90	3.01	1.20	1.39	5.51	6.39
2	Be	24.2	1.88	1.93	2.23	14.14	16.41
	Mg	8.60	2.65	1.37	1.58	7.13	8.27
	Ca	4.60	3.27	1.11	1.28	4.68	5.43
	Sr	3.56	3.56	1.02	1.18	3.95	4.58
	Ba	3.20	3.69	0.98	1.13	3.65	4.24
	Zn	13.10	2.31	1.57	1.82	9.39	10.90
	Cd	9.28	2.59	1.40	1.62	7.46	8.66
3	Al	18.06	2.07	1.75	2.02	11.63	13.49
	Ga	15.30	2.19	1.65	1.91	10.35	12.01
	In	11.49	2.41	1.50	1.74	8.60	9.98
4	Pb	13.20	2.30	1.57	1.82	9.37	10.87
	Sn( <i>w</i> )	14.48	2.23	1.62	1.88	10.03	11.64

<sup>a</sup>The dimensionless radius parameter is defined as  $r_n = r_0/a_H$ , where  $a_H$  is the first Bohr radius and  $r_0$  is the radius of a sphere that contains one electron.

$$k_F = (3\pi^2 n)^{1/3}$$

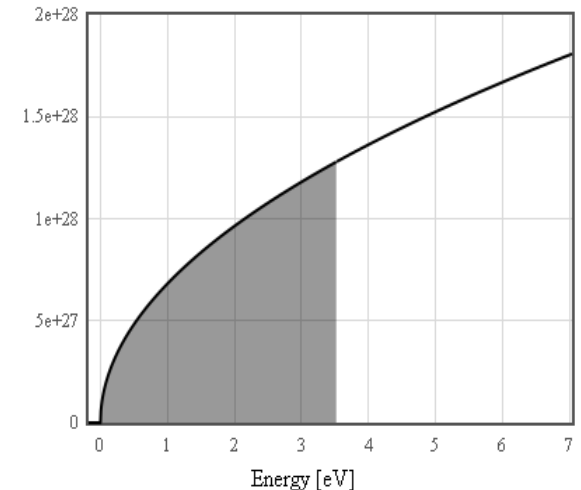
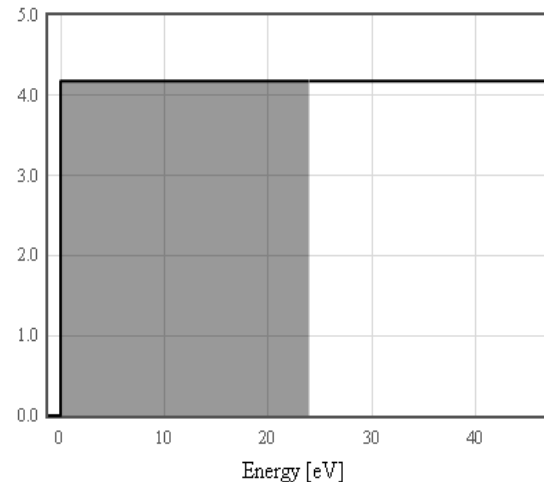
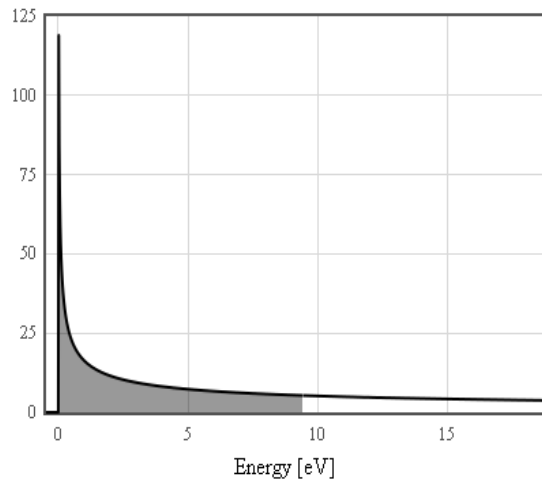
$$E_F \gg k_B T$$

# Free electron Fermi gas

$$1 - d \quad D(E) = \sqrt{\frac{2m}{\hbar^2 \pi^2 E}} = \frac{n}{2\sqrt{E_F E}} \quad \text{J}^{-1} \text{m}^{-1}$$

$$2 - d \quad D(E) = \frac{m}{\hbar^2 \pi} = \frac{n}{E_F} \quad \text{J}^{-1} \text{m}^{-2}$$

$$3 - d \quad D(E) = \frac{\pi}{2} \left( \frac{2m}{\hbar^2 \pi^2} \right)^{3/2} \sqrt{E} = \frac{3n}{2E_F^{3/2}} \sqrt{E} \quad \text{J}^{-1} \text{m}^{-3}$$



# Average electron energy

---

$$n\langle E \rangle = \int_0^{E_F} ED(E)dE$$

$$D(E) = \frac{(2m)^{\frac{3}{2}}}{2\pi^2\hbar^3} \sqrt{E} = \frac{3n}{2E_F^{3/2}} \sqrt{E} \quad \text{J}^{-1}\text{m}^{-3}$$

$$n\langle E \rangle = \int_0^{E_F} \frac{3n}{2E_F^{3/2}} E^{\frac{3}{2}} dE = \frac{3}{5} nE_F$$

$$\langle E \rangle = \frac{3}{5} E_F$$

$$u(T=0) = \frac{3}{5} nE_F$$

$$E_F = \frac{\hbar^2}{2m} (3\pi^2 n)^{2/3}$$

$$u(T=0) = \frac{\pi^{\frac{4}{3}} \hbar^2}{10m} (3n)^{\frac{5}{3}} = \frac{\pi^{\frac{4}{3}} \hbar^2}{10m} \left( \frac{3N}{V} \right)^{\frac{5}{3}}$$

# Pressure 3-D

---

$$P = - \left( \frac{\partial U}{\partial V} \right)_N$$

$$u(T = 0) = \frac{\pi^{\frac{4}{3}} \hbar^2}{10m} (3n)^{\frac{5}{3}} = \frac{\pi^{\frac{4}{3}} \hbar^2}{10m} \left( \frac{3N}{V} \right)^{\frac{5}{3}}$$

$$U = Vu \propto V^{-2/3}$$

$$P = - \left( \frac{\partial U}{\partial V} \right)_N = \frac{2}{3} \frac{U}{V} = \frac{2}{5} n E_F = \frac{\hbar^2 (9\pi^4 n^5)^{\frac{1}{3}}}{5m}$$

# Bulk modulus

---

$$B = -V \frac{\partial P}{\partial V}$$

$$P = - \left( \frac{\partial U}{\partial V} \right)_N = \frac{\hbar^2 (9\pi^4 N^5 / V^5)^{\frac{1}{3}}}{5m}$$

$$P \propto V^{-5/3}$$

$$B = \frac{5}{3} P = \frac{10 U}{9 V} = \frac{2}{3} n E_F = \frac{\hbar^2 (3\pi^4 n^5)^{\frac{1}{3}}}{m} \quad \text{N/m}^2$$

See: Landau and Lifshitz, Statistical Physics 1  
or Ashcroft and Mermin, Solid State Physics

# Bulk modulus

---

Table 2.2

**BULK MODULI IN  $10^{10}$  DYNES/CM<sup>2</sup> FOR SOME TYPICAL METALS<sup>a</sup>**

METAL	FREE ELECTRON $B$	MEASURED $B$
Li	23.9	11.5
Na	9.23	6.42
K	3.19	2.81
Rb	2.28	1.92
Cs	1.54	1.43
Cu	63.8	134.3
Ag	34.5	99.9
Al	228	76.0

<sup>a</sup> The free electron value is that for a free electron gas at the observed density of the metal, as calculated from Eq. (2.37).

---

## Results of the quantization of the Schrödinger equation for free fermions in 1, 2, and 3 dimensions.

A simple model for metals is the free electron model where the potential energy of the electrons is zero and the electron-electron interactions are ignored. This is equivalent to any system of noninteracting fermions with zero potential energy. In this model the thermodynamic properties only depend on one parameter, the particle density  $n$ . In the table below,  $n$  denotes the number of particles per meter in one-dimension, the number of particles per square meter in two-dimensions, and the number of particles per cubic meter in three dimensions.

	1-D Schrödinger equation for a free particle	2-D Schrödinger equation for a free particle	3-D Schrödinger equation for a free particle
	$i\hbar \frac{d\psi}{dx} = -\frac{\hbar^2}{2m} \frac{d^2\psi}{dx^2}$	$i\hbar \frac{d\psi}{dx} = -\frac{\hbar^2}{2m} \left( \frac{d^2\psi}{dx^2} + \frac{d^2\psi}{dy^2} \right)$	$i\hbar \frac{d\psi}{dx} = -\frac{\hbar^2}{2m} \left( \frac{d^2\psi}{dx^2} + \frac{d^2\psi}{dy^2} + \frac{d^2\psi}{dz^2} \right)$
Eigenfunction solutions	$A_k \exp(i(kx - \alpha t))$	$A_{\vec{k}} \exp(i(\vec{k} \cdot \vec{r} - \alpha t))$	$A_{\vec{k}} \exp(i(\vec{k} \cdot \vec{r} - \alpha t))$
Dispersion relation	$E = \hbar\omega = \frac{\hbar^2 k^2}{2m} \quad \text{J}$	$E = \hbar\omega = \frac{\hbar^2 k^2}{2m} \quad \text{J}$	$E = \hbar\omega = \frac{\hbar^2 k^2}{2m} \quad \text{J}$
Density of states	$D(k) = \frac{2}{\pi}$	$D(k) = \frac{k}{\pi} \quad \text{m}^{-1}$	$D(k) = \frac{k^2}{\pi^2} \quad \text{m}^{-2}$
Density of states $D(E) = D(k) \frac{dk}{dE}$	$D(E) = \frac{1}{\pi\hbar} \sqrt{\frac{2m}{E}} = \frac{n}{2\sqrt{E_F E}} \quad \text{J}^{-1}\text{m}^{-1}$	$D(E) = \frac{m}{\pi\hbar^2} = \frac{n}{E_F} \quad \text{J}^{-1}\text{m}^{-2}$	$D(E) = \frac{(2m)^{3/2}}{2\pi^2\hbar^3} \sqrt{E} = \frac{3n}{2E_F^{3/2}} \sqrt{E} \quad \text{J}^{-1}\text{m}^{-3}$
Fermi energy $E_F$ $n = \int_{-\infty}^{E_F} D(E) dE$	$E_F = \frac{\pi^2 \hbar^2 n^2}{8m} \quad \text{J}$	$E_F = \frac{\pi \hbar^2 n}{m} \quad \text{J}$	$E_F = \frac{\hbar^2}{2m} (3\pi^2 n)^{2/3} \quad \text{J}$
$D(E_F)$	$D(E_F) = \frac{4m}{\pi^2 \hbar^2 n} \quad \text{J}^{-1}\text{m}^{-1}$	$D(E_F) = \frac{m}{\pi \hbar^2} \quad \text{J}^{-1}\text{m}^{-2}$	$D(E_F) = \frac{m(3n)^{1/3}}{4\pi^3 \hbar^2} \quad \text{J}^{-1}\text{m}^{-3}$
$D'(E_F) = \left. \frac{dD}{dE} \right _{E=E_F}$	$D'(E_F) = \frac{-16m^2}{\pi^4 \hbar^4 n^3} \quad \text{J}^{-2}\text{m}^{-1}$	$D'(E_F) = 0 \quad \text{J}^{-2}\text{m}^{-2}$	$D'(E_F) = \frac{m^2}{\hbar^4 \sqrt[3]{3\pi^8 n}} \quad \text{J}^{-2}\text{m}^{-3}$
Chemical potential $\mu$ $n = \int_{-\infty}^{\mu} D(E) f(E) dE$	$\mu \approx E_F - \frac{\pi^2}{6} (k_B T)^2 \frac{D'(E_F)}{D(E_F)} \quad \text{J}$ $\approx \frac{\pi^2 \hbar^2 n^2}{8m} + \frac{2m}{3\hbar^2 n^2} (k_B T)^2 \quad \text{J}$	$\mu = k_B T \ln \left( \exp \left( \frac{E_F}{k_B T} \right) - 1 \right) \quad \text{J}$ $= k_B T \ln \left( \exp \left( \frac{\pi \hbar^2 n}{m k_B T} \right) - 1 \right) \quad \text{J}$	$\mu \approx E_F - \frac{\pi^2}{6} (k_B T)^2 \frac{D'(E_F)}{D(E_F)} \quad \text{J}$ $\approx \frac{\hbar^2}{2m} (3\pi^2 n)^{2/3} - \frac{\pi^3 m}{10 \sqrt[3]{3\pi^8 n}} (k_B T)^2 \quad \text{J}$





Arnold  
Sommerfeld

# Sommerfeld Expansion

We would like to perform integrals of the form

$$\int_{-\infty}^{\infty} H(E) f(E) dE$$

Examples:

$$n = \int_{-\infty}^{\infty} D(E) f(E) dE \quad u = \int_{-\infty}^{\infty} ED(E) f(E) dE$$

Integrate by parts (Partielle Integration)

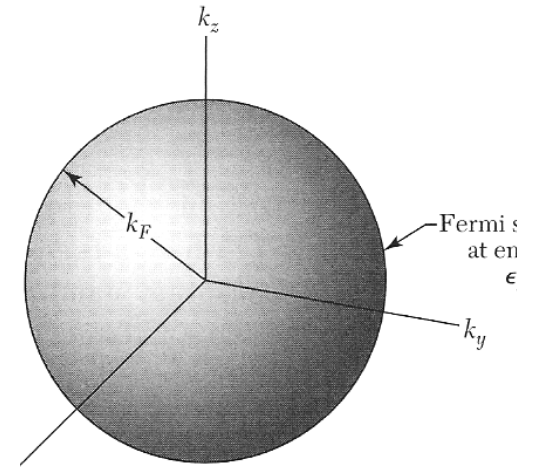
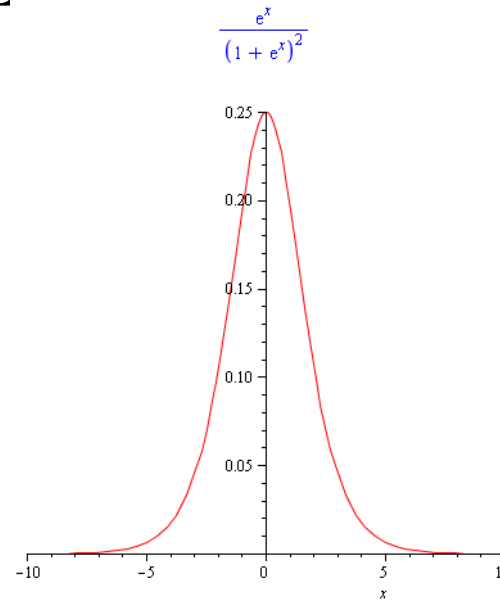
$$\int_{-\infty}^{\infty} \frac{dK(E)}{dE} f(E) dE = K(\infty) f(\infty) - K(-\infty) f(-\infty) - \int_{-\infty}^{\infty} K(E) \frac{df(E)}{dE} dE$$

$$K(E) = \int_{-\infty}^E H(E') dE' \quad H(E) = \frac{dK(E)}{dE}$$

# Sommerfeld Expansion

$$\int_{-\infty}^{\infty} H(E) f(E) dE = - \int_{-\infty}^{\infty} K(E) \frac{df(E)}{dE} dE$$

$$\frac{-df(E)}{dE} = \frac{\frac{1}{k_B T} \exp\left(\frac{E - \mu}{k_B T}\right)}{\left(1 + \exp\left(\frac{E - \mu}{k_B T}\right)\right)^2}$$



Expand  $K(E)$  around  $E = \mu$

$$K(E) \approx K(\mu) + \left. \frac{dK}{dE} \right|_{E=\mu} (E - \mu) + \frac{1}{2} \left. \frac{d^2 K}{dE^2} \right|_{E=\mu} (E - \mu)^2 + \dots$$

# Sommerfeld Expansion

---

$$x = \frac{E - \mu}{k_B T}$$

$$\int_{-\infty}^{\infty} \frac{e^x}{(1 + e^x)^2} dx = 1$$

$$\int_{-\infty}^{\infty} \frac{x e^x}{(1 + e^x)^2} dx = 0$$

$$\int_{-\infty}^{\infty} \frac{x^2 e^x}{(1 + e^x)^2} dx = \frac{\pi^2}{3}$$

$$\int_{-\infty}^{\infty} \frac{x^3 e^x}{(1 + e^x)^2} dx = 0$$

$$\int_{-\infty}^{\infty} \frac{x^4 e^x}{(1 + e^x)^2} dx = \frac{7\pi^4}{15}$$

# Sommerfeld Expansion

---

$$\int_{-\infty}^{\infty} H(E) f(E) dE \approx K(\mu) + \frac{\pi^2}{6} (k_B T)^2 \left. \frac{dH(E)}{dE} \right|_{E=\mu} + \frac{7\pi^4}{360} (k_B T)^4 \left. \frac{d^3 H(E)}{dE^3} \right|_{E=\mu} + \dots$$

$$K(\mu) = \int_{-\infty}^{\mu} H(E) dE \qquad H(E) = \frac{dK(E)}{dE}$$

# Sommerfeld Expansion: chemical potential

---

$$\int_{-\infty}^{\infty} H(E) f(E) dE = K(\mu) + \frac{\pi^2}{6} (k_B T)^2 \left. \frac{dH(E)}{dE} \right|_{E=\mu} + \frac{7\pi^4}{360} (k_B T)^4 \left. \frac{d^3 H(E)}{dE^3} \right|_{E=\mu} + \dots$$

For  $\mu$  in 3-d:

$$n = \int_{-\infty}^{\infty} D(E) f(E) dE$$

$$n = \int_{-\infty}^{\mu} D(E) dE + \frac{\pi^2}{6} (k_B T)^2 D'(E_F) + \dots$$

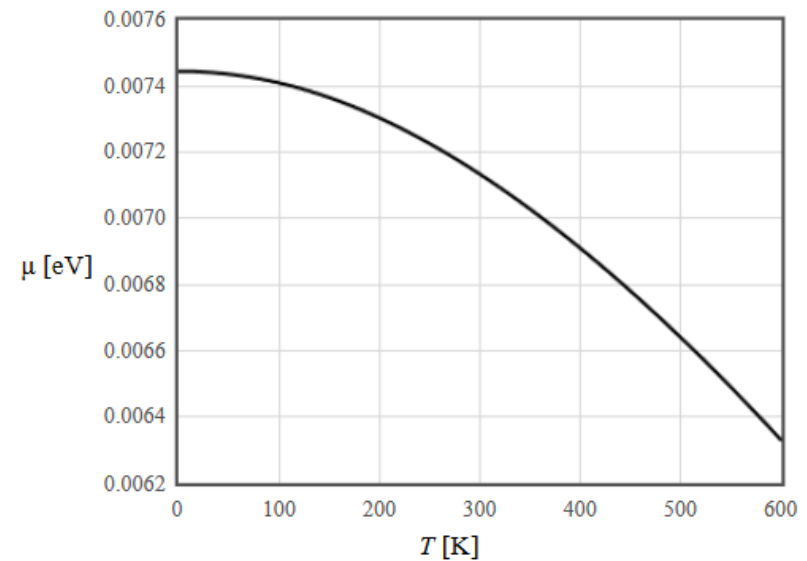
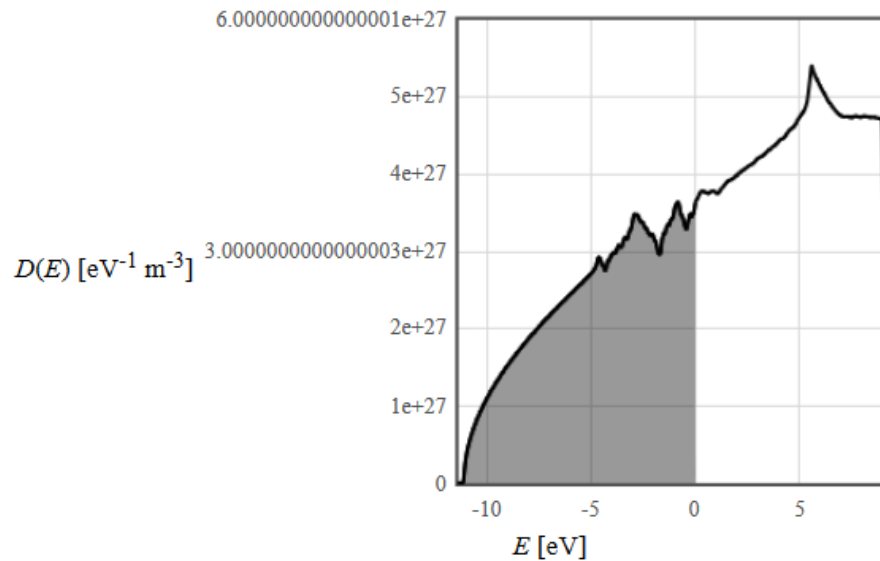


$$K(\mu) = \int_{-\infty}^{\mu} D(E) dE = \int_{-\infty}^{E_F} D(E) dE + \int_{E_F}^{\mu} D(E) dE \approx n + (\mu - E_F) D(E_F)$$

# Sommerfeld Expansion: chemical potential

$$n = n + (\mu - E_F) D(E_F) + \frac{\pi^2}{6} (k_B T)^2 D'(E_F)$$

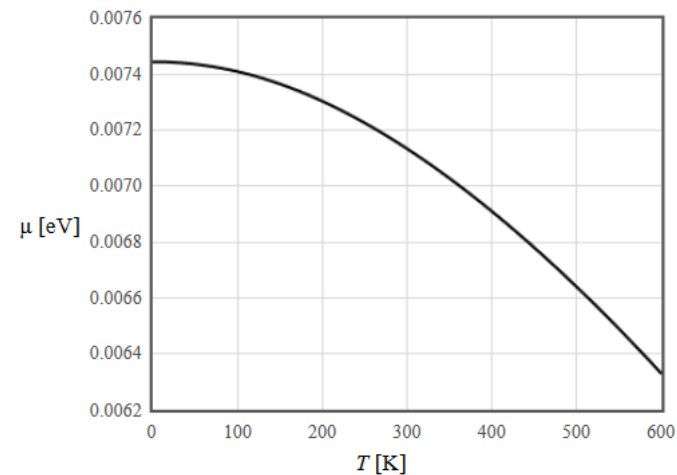
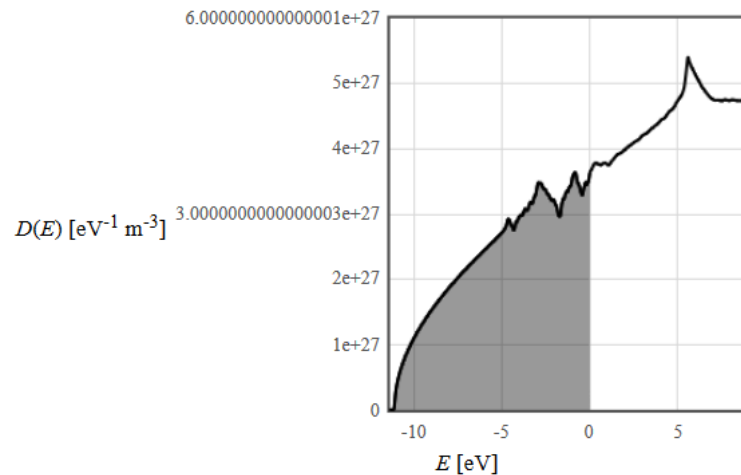
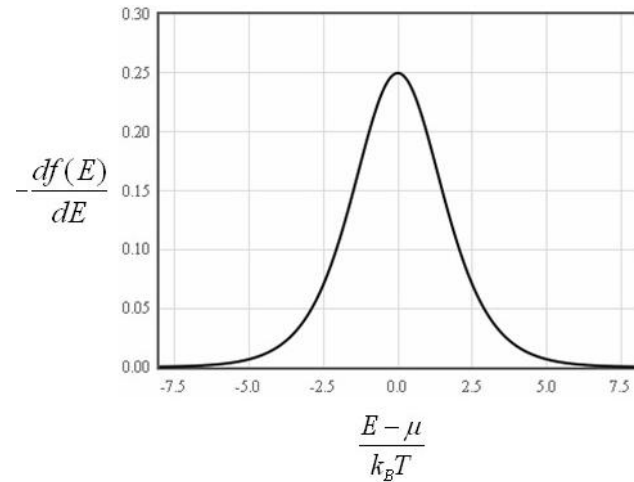
$$\mu \approx E_F - \frac{\pi^2}{6} (k_B T)^2 \frac{D'(E_F)}{D(E_F)}$$



Aluminum

# Sommerfeld Expansion: chemical potential

$$n = - \int_{-\infty}^{\infty} K(E) \frac{df(E)}{dE} dE = \int_{-\infty}^{\infty} \frac{K(E) \exp\left(\frac{E - \mu}{k_B T}\right) dE}{k_B T \left( \exp\left(\frac{E - \mu}{k_B T}\right) + 1 \right)^2}$$



<http://lampx.tugraz.at/~hadley/ss1/materials/thermo/dos2mu.html>