

20. Electron Bands

June 7, 2018

Bloch Theorem

$$\psi_{\vec{k}}(\vec{r}) = e^{i\vec{k}\cdot\vec{r}} u_{\vec{k}}(\vec{r})$$

$$\psi(\vec{r}) = \sum_{\vec{k}} C_{\vec{k}} e^{i\vec{k}\cdot\vec{r}} \leftarrow \text{Any wave function that satisfies periodic boundary conditions}$$

$$\psi(\vec{r}) = \sum_{\vec{k} \in 1Bz} \sum_{\vec{G}} C_{\vec{k}+\vec{G}} e^{i(\vec{k}+\vec{G})\cdot\vec{r}}$$

These k 's label the symmetries \nearrow

periodic function \swarrow

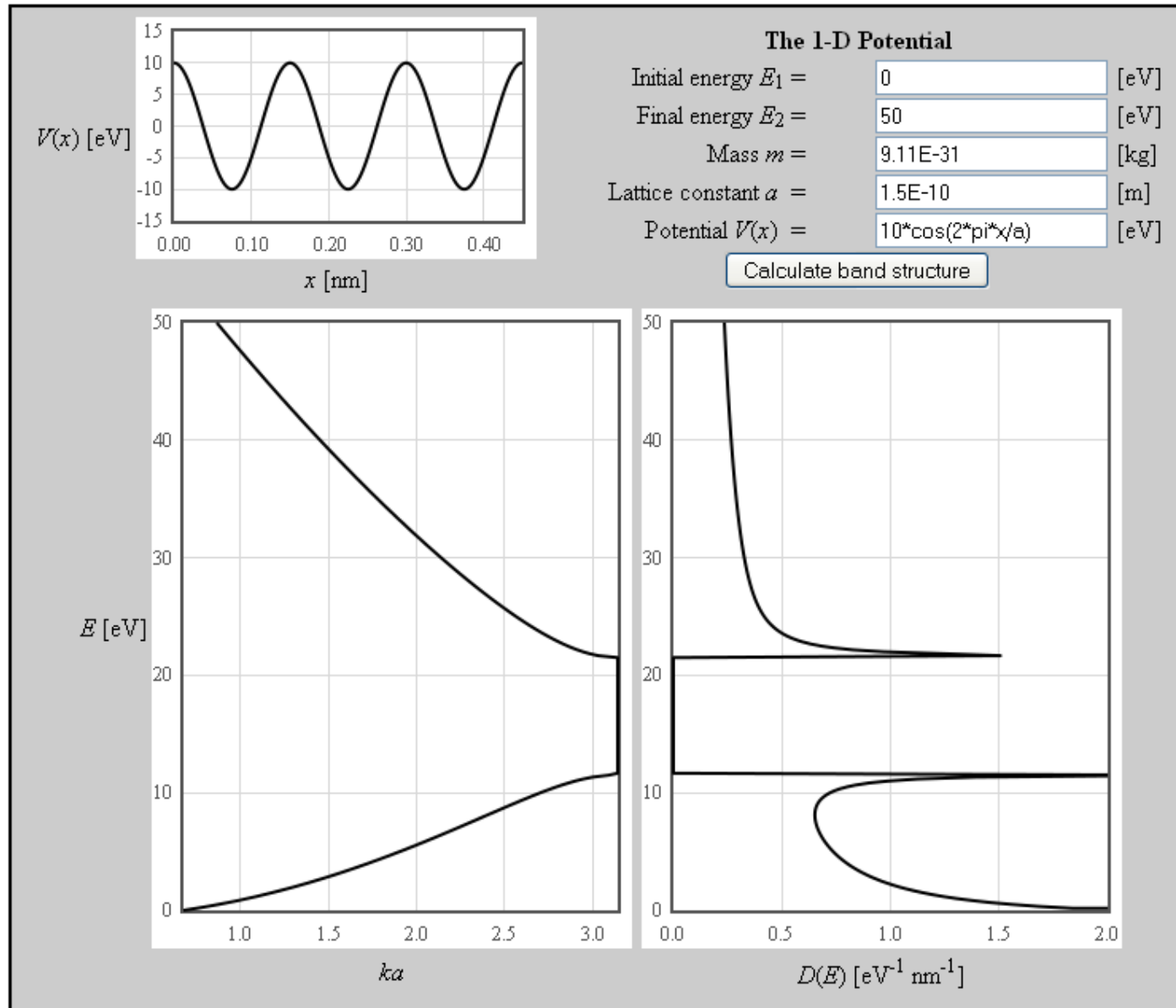
$$\psi_{\vec{k}}(\vec{r}) = \sum_{\vec{G}} C_{\vec{k}+\vec{G}} e^{i(\vec{k}+\vec{G})\cdot\vec{r}} = e^{i\vec{k}\cdot\vec{r}} \sum_{\vec{G}} C_{\vec{k}+\vec{G}} e^{i\vec{G}\cdot\vec{r}} = e^{i\vec{k}\cdot\vec{r}} u_{\vec{k}}(\vec{r})$$

$$T_{mnl} \psi_{\vec{k}}(\vec{r}) = e^{i\vec{k}\cdot(\vec{r}+m\vec{a}_1+n\vec{a}_2+l\vec{a}_3)} u_{\vec{k}}(\vec{r}+m\vec{a}_1+n\vec{a}_2+l\vec{a}_3) = e^{i\vec{k}\cdot(m\vec{a}_1+n\vec{a}_2+l\vec{a}_3)} \psi_{\vec{k}}(\vec{r})$$

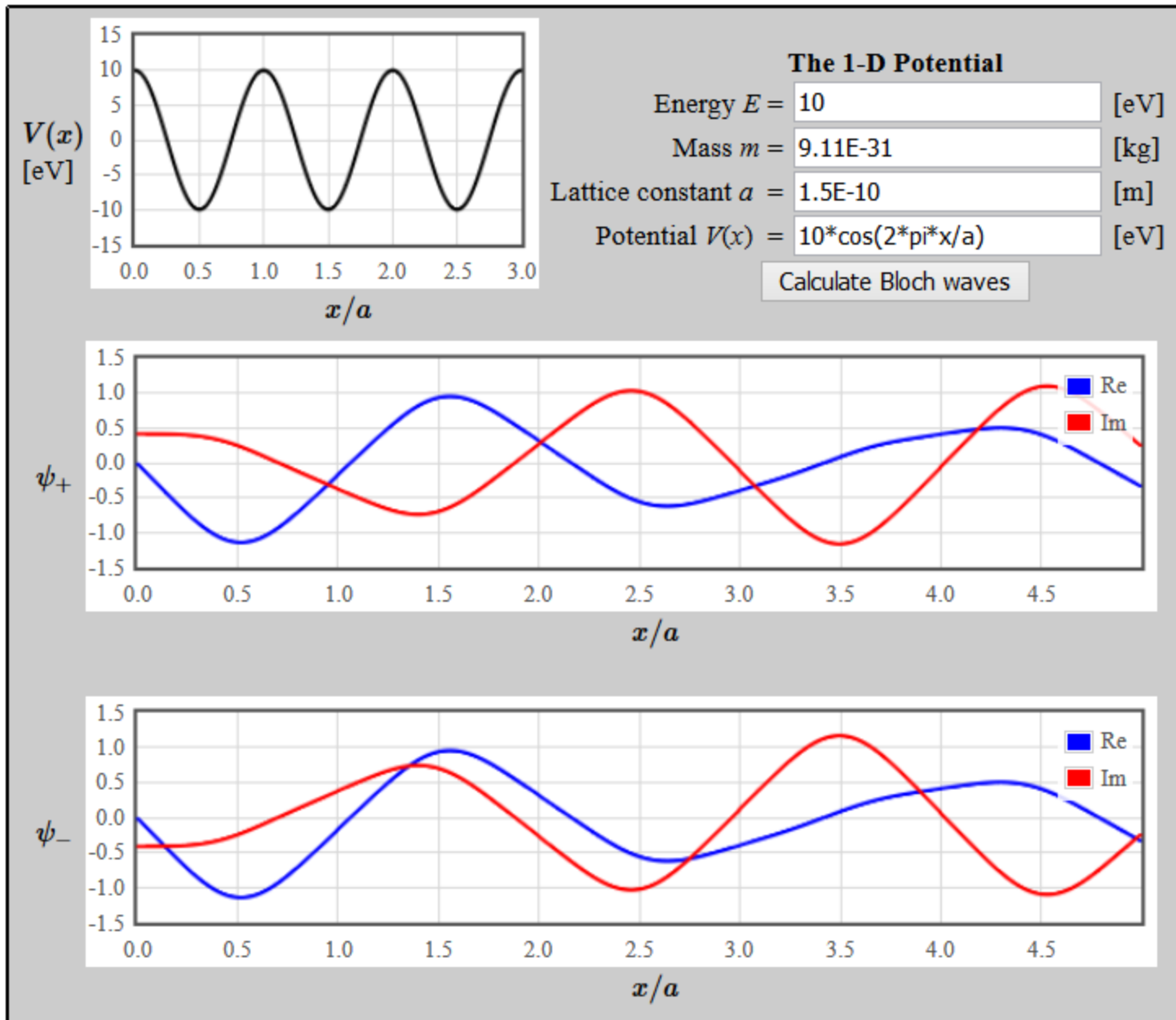
Bloch form $\psi_{\vec{k}}(\vec{r}) = e^{i\vec{k}\cdot\vec{r}} u_{\vec{k}}(\vec{r})$

Eigen function solutions of the Schrödinger equation have Bloch form.

Band structure in 1-D



Bloch waves in 1-D



Plane wave method

$$-\frac{\hbar^2}{2m} \nabla^2 \psi + U_{MO}(\vec{r}) \psi = E \psi$$

Write U and ψ as Fourier series.

$$U_{MO}(\vec{r}) = \sum_{\vec{G}} U_{\vec{G}} e^{i\vec{G}\cdot\vec{r}} \quad \psi(\vec{r}) = \sum_{\vec{k}} C_{\vec{k}} e^{i\vec{k}\cdot\vec{r}}$$

For the molecular orbital Hamiltonian

$$U_{MO}(\vec{r}) = \frac{-Ze^2}{4\pi\epsilon_0} \sum_j \frac{1}{|\vec{r} - \vec{r}_j|} = \frac{-Ze^2}{V\epsilon_0} \sum_{\vec{G}} \frac{e^{i\vec{G}\cdot\vec{r}}}{G^2}$$

volume of a unit cell



Plane wave method

$$-\frac{\hbar^2}{2m}\nabla^2\psi + U(\vec{r})\psi = E\psi$$

$$U_{MO}(\vec{r}) = \sum_{\vec{G}} U_{\vec{G}} e^{i\vec{G}\cdot\vec{r}} \quad \psi(\vec{r}) = \sum_{\vec{k}} C_{\vec{k}} e^{i\vec{k}\cdot\vec{r}}$$

$$\sum_{\vec{k}} \frac{\hbar^2 k^2}{2m} C_{\vec{k}} e^{i\vec{k}\cdot\vec{r}} + \sum_{\vec{G}} \sum_{\vec{k}'} U_{\vec{G}} C_{\vec{k}'} e^{i(\vec{G}+\vec{k}')\cdot\vec{r}} = E \sum_{\vec{k}} C_{\vec{k}} e^{i\vec{k}\cdot\vec{r}}$$

Must hold for each Fourier coefficient.

$$\vec{k}' + \vec{G} = \vec{k} \quad \Rightarrow \quad \vec{k}' = \vec{k} - \vec{G}$$

$$\left(\frac{\hbar^2 k^2}{2m} - E \right) C_{\vec{k}} + \sum_{\vec{G}} U_{\vec{G}} C_{\vec{k}-\vec{G}} = 0$$

Central equations (one for every k in the first Brillouin zone)

Plane wave method

The central equations can be written as a matrix equation.

$$M\vec{C} = E\vec{C}$$

Diagonal elements: $M_{ii} = \frac{\hbar^2}{2m} (\vec{k} - \vec{G}_i)^2$

Off-diagonal elements: $M_{ij} = -\frac{Ze^2}{V\epsilon_0 (\vec{G}_i - \vec{G}_j)^2}$

Central equations - one dimension

$$\left(\frac{\hbar^2 k^2}{2m} - E \right) C_k + \sum_G U_G C_{k-G} = 0$$

$$\begin{bmatrix} \dots & & & & & & & & & \\ \frac{\hbar^2 (k - 2G_0)^2}{2m} - E & U_{G_0} & U_{2G_0} & U_{3G_0} & U_{4G_0} & U_{5G_0} & & & & \\ U_{-G_0} & \frac{\hbar^2 (k - G_0)^2}{2m} - E & U_{G_0} & U_{2G_0} & U_{3G_0} & U_{4G_0} & & & & \\ U_{-2G_0} & U_{-G_0} & \frac{\hbar^2 k^2}{2m} - E & U_{G_0} & U_{2G_0} & U_{3G_0} & & & & \\ U_{-3G_0} & U_{-2G_0} & U_{-G_0} & \frac{\hbar^2 (k + G_0)^2}{2m} - E & U_{G_0} & U_{2G_0} & & & & \\ U_{-4G_0} & U_{-3G_0} & U_{-2G_0} & U_{-G_0} & \frac{\hbar^2 (k + 2G_0)^2}{2m} - E & U_{G_0} & & & & \\ & & & & & \dots & & & & \end{bmatrix} \begin{bmatrix} \vdots \\ C_{k+2G_0} \\ C_{k+G_0} \\ C_k \\ C_{k-G_0} \\ C_{k-2G_0} \\ \vdots \end{bmatrix} = 0$$

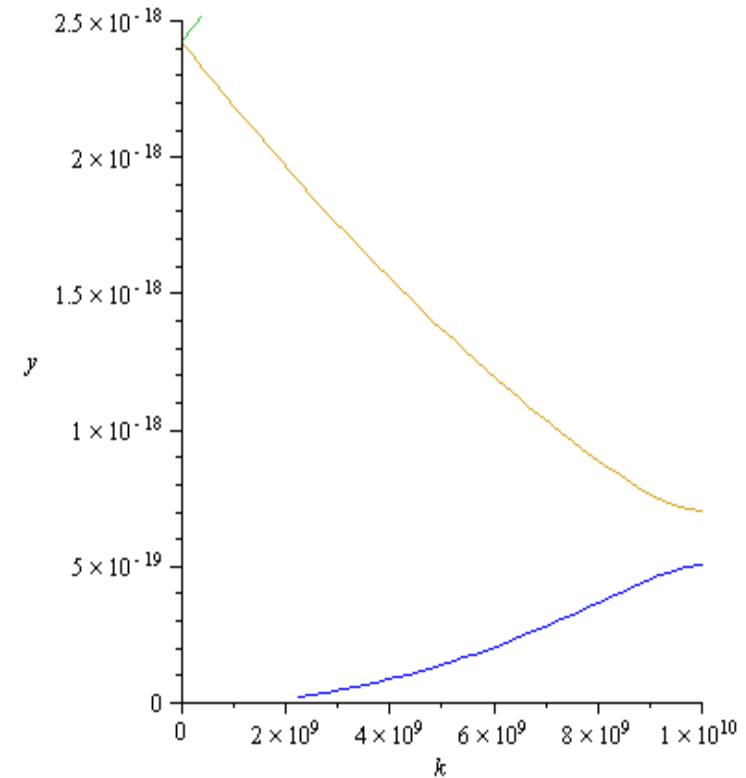
Central equations couple coefficients k to other coefficients that differ by a reciprocal lattice wavevector G .

Central equations - one dimension

$$M4 := \begin{bmatrix} \frac{\hbar^2 \cdot (k + 2 \cdot G)^2}{2m} & U & 0 & 0 \\ U & \frac{\hbar^2 \cdot (k + G)^2}{2m} & U & 0 \\ 0 & U & \frac{\hbar^2 \cdot k^2}{2m} & U \\ 0 & 0 & U & \frac{\hbar^2 \cdot (k - G)^2}{2m} \end{bmatrix} :$$

$V4 := \text{Eigenvalues}(M4) :$

$\text{plot}([V4[1], V4[2], V4[3], V4[4]], k = 0 \dots 1E10, y = 0 \dots 2.5E-18);$



Central equations 3d - simple cubic

$$V(\vec{r}) = \sum_{\vec{G}} U_{\vec{G}} e^{i\vec{G}\cdot\vec{r}}$$

Molecular orbital Hamiltonian

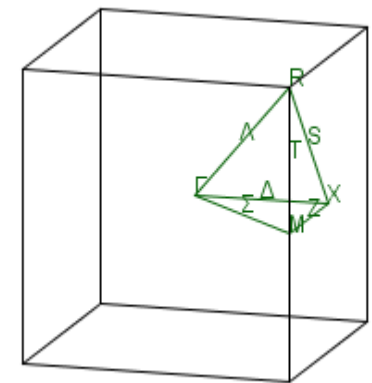
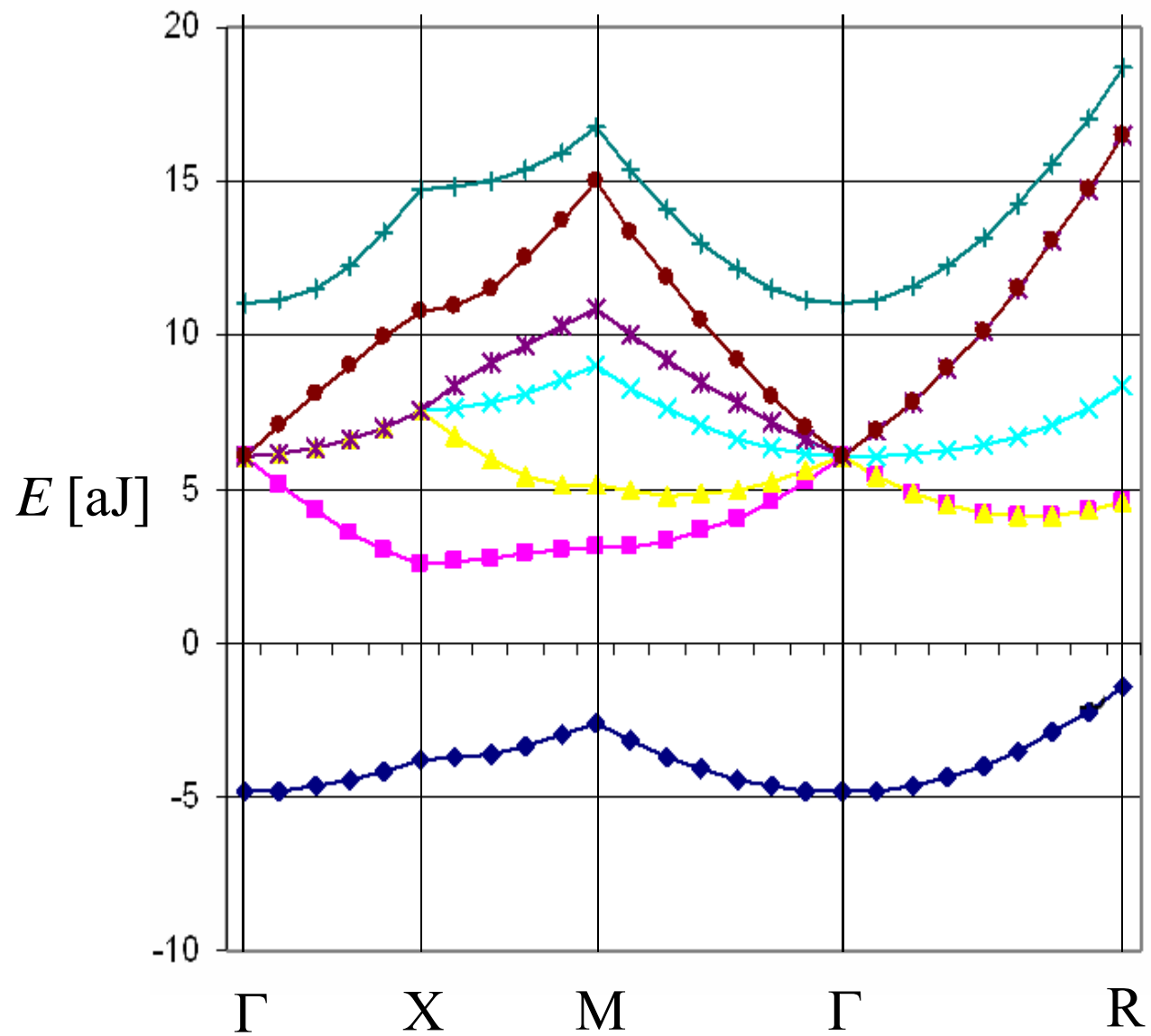
$$U_{\vec{G}} = \frac{-Ze^2}{V_{\text{unit cell}} \epsilon_0 G^2}$$

Central equations:
$$\left(\frac{\hbar^2 k^2}{2m} - E \right) C_{\vec{k}} + \sum_{\vec{G}} U_{\vec{G}} C_{\vec{k}-\vec{G}} = 0$$

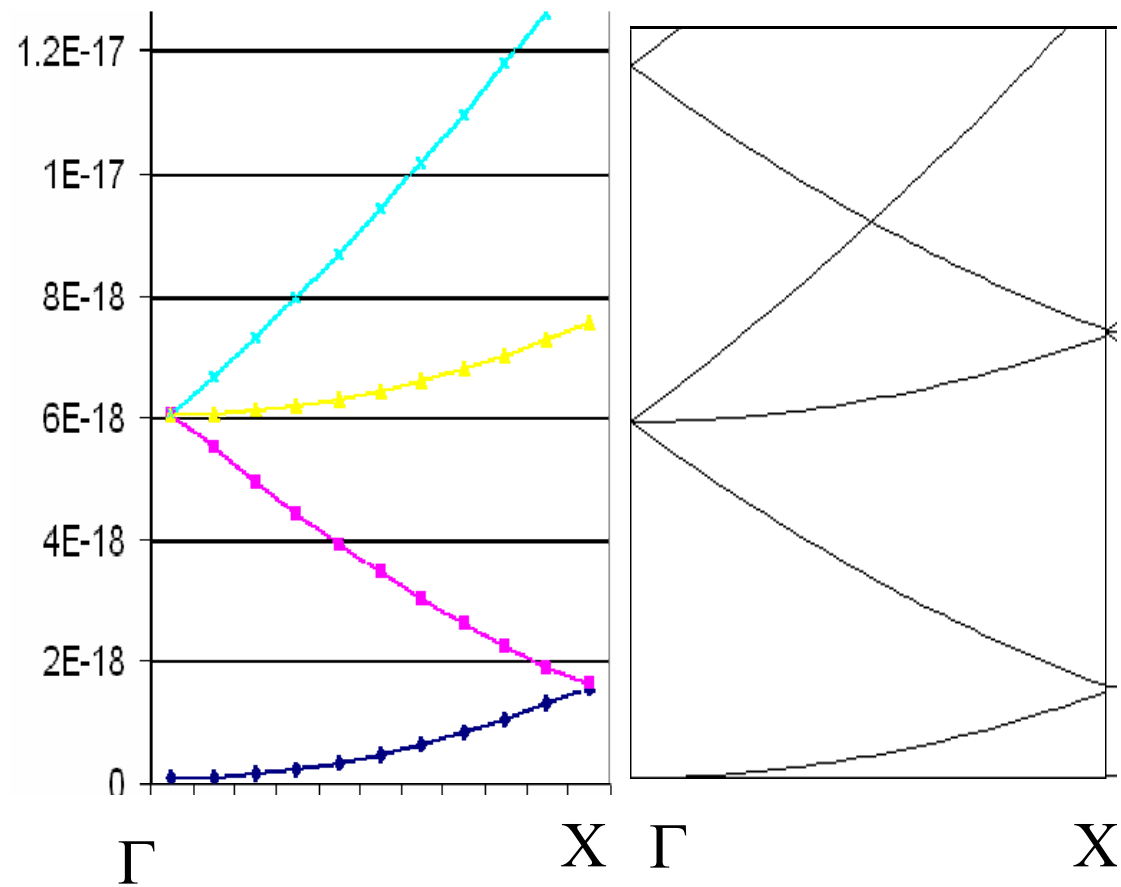
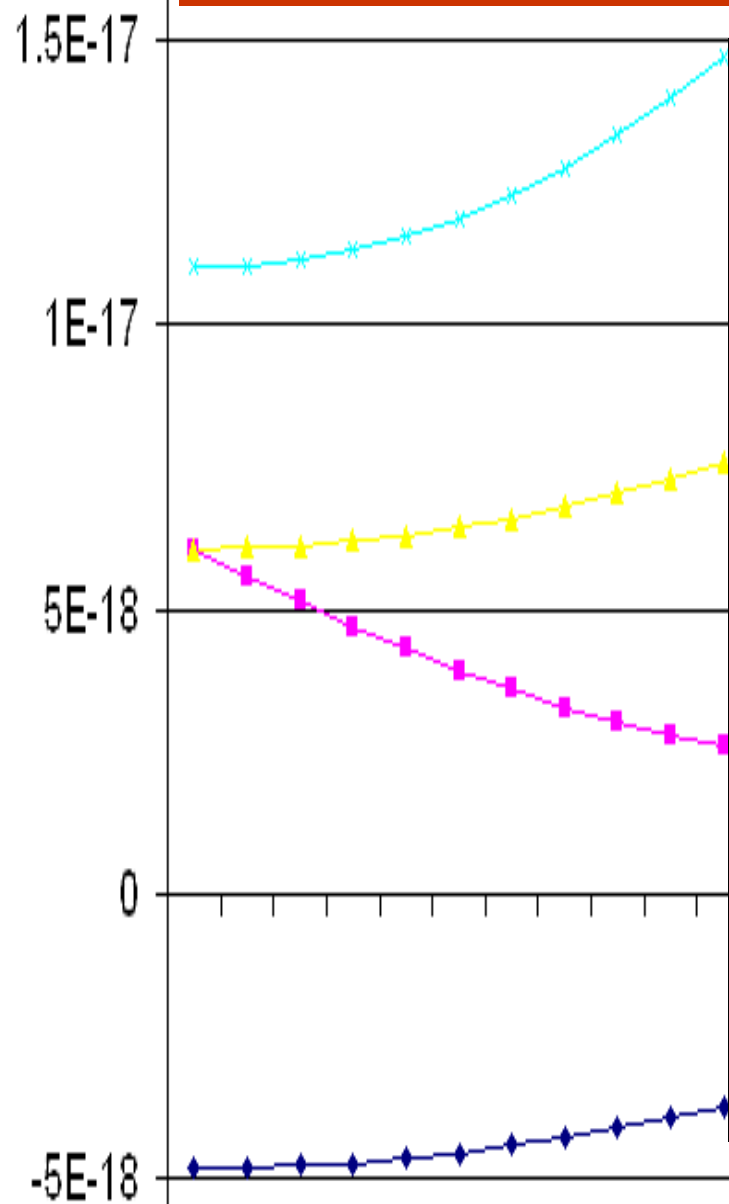
diagonal elements:
$$\frac{\hbar^2}{2m} (\vec{k} - \vec{G}_i)^2$$

off-diagonal elements:
$$\frac{-Ze^2}{V_{\text{unit cell}} \epsilon_0 (\vec{G}_i - \vec{G}_j)^2}$$

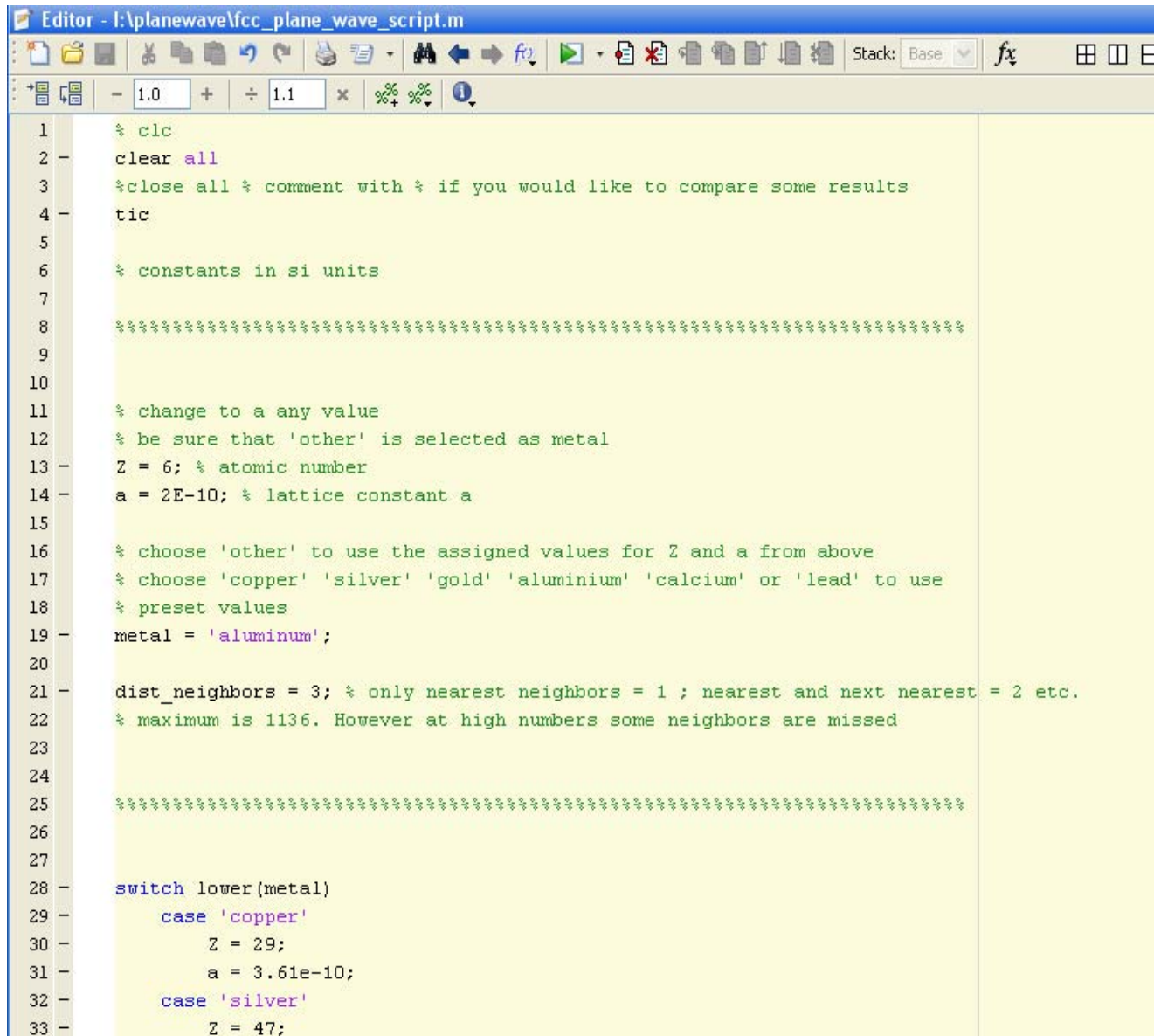
Central equations - simple cubic



Central equations - simple cubic



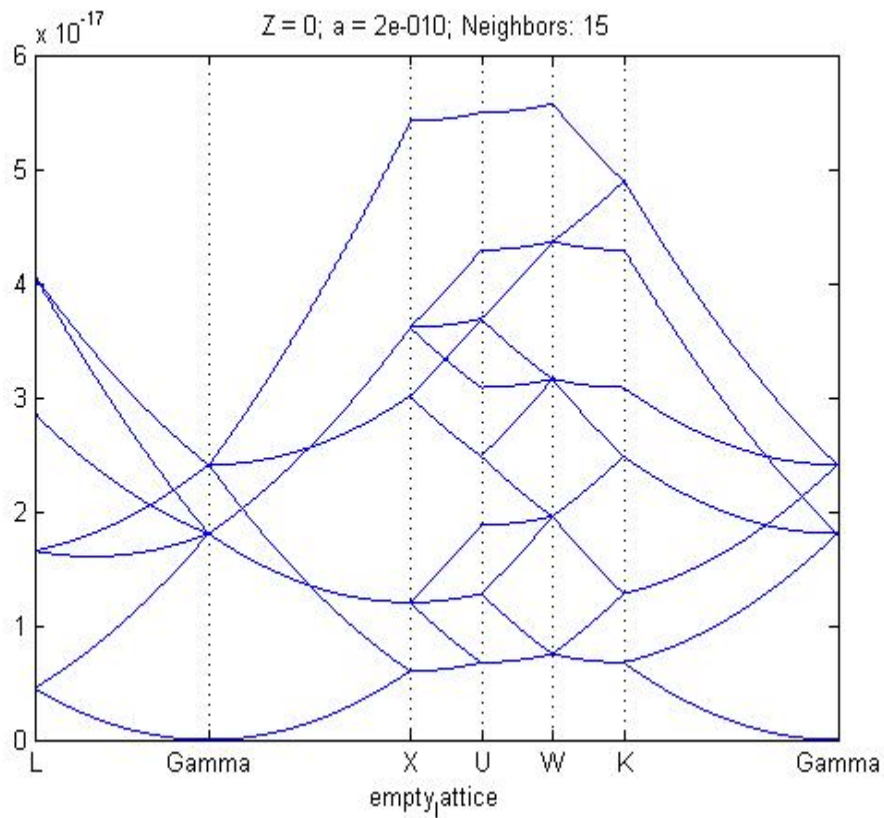
Plane wave method



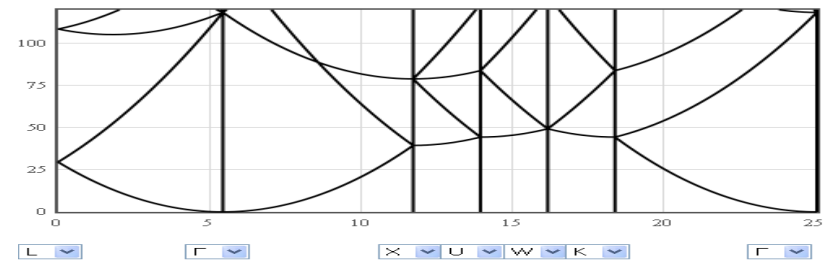
```
1  % clc
2  clear all
3  %close all % comment with % if you would like to compare some results
4  tic
5
6  % constants in si units
7
8  %*****
9
10
11 % change to a any value
12 % be sure that 'other' is selected as metal
13 Z = 6; % atomic number
14 a = 2E-10; % lattice constant a
15
16 % choose 'other' to use the assigned values for Z and a from above
17 % choose 'copper' 'silver' 'gold' 'aluminium' 'calcium' or 'lead' to use
18 % preset values
19 metal = 'aluminum';
20
21 dist_neighbors = 3; % only nearest neighbors = 1 ; nearest and next nearest = 2 etc.
22 % maximum is 1136. However at high numbers some neighbors are missed
23
24
25 %*****
26
27
28 switch lower(metal)
29     case 'copper'
30         Z = 29;
31         a = 3.61e-10;
32     case 'silver'
33         Z = 47;
```

Plane wave method

fcc $Z=0$

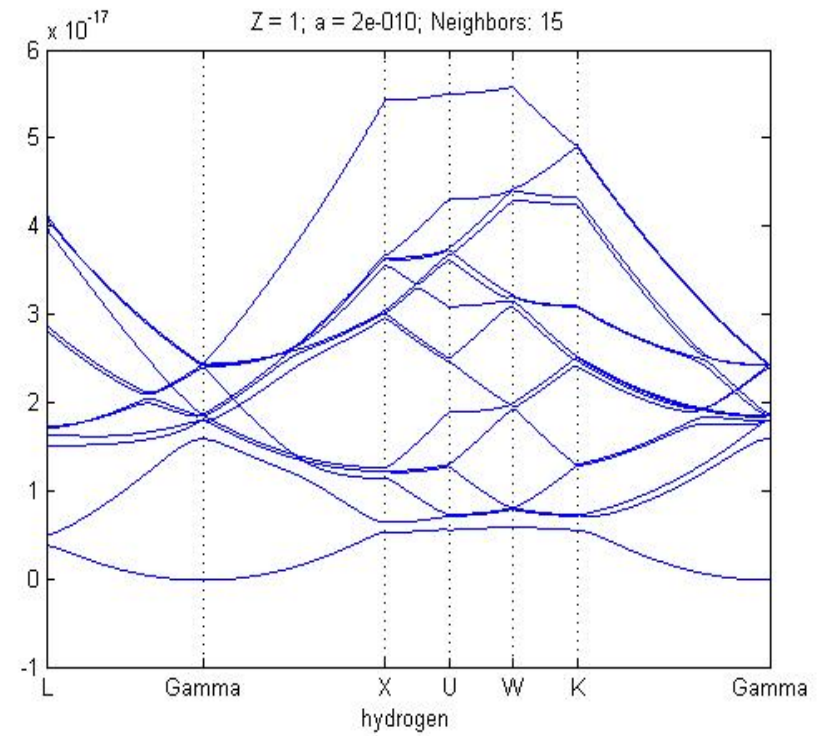
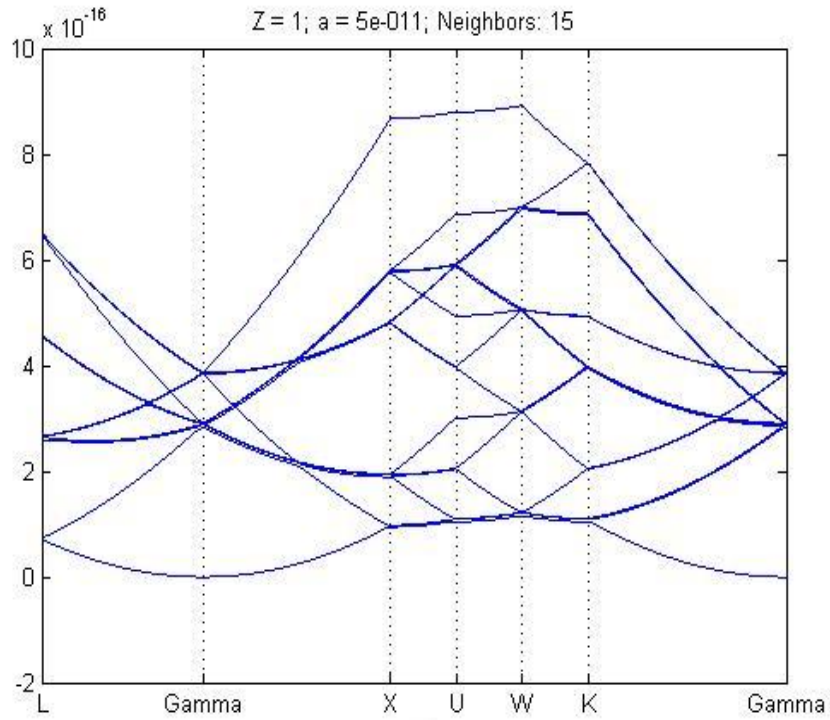


empty lattice



Plane wave method

fcc hydrogen



Approximate solution near the Bz boundary

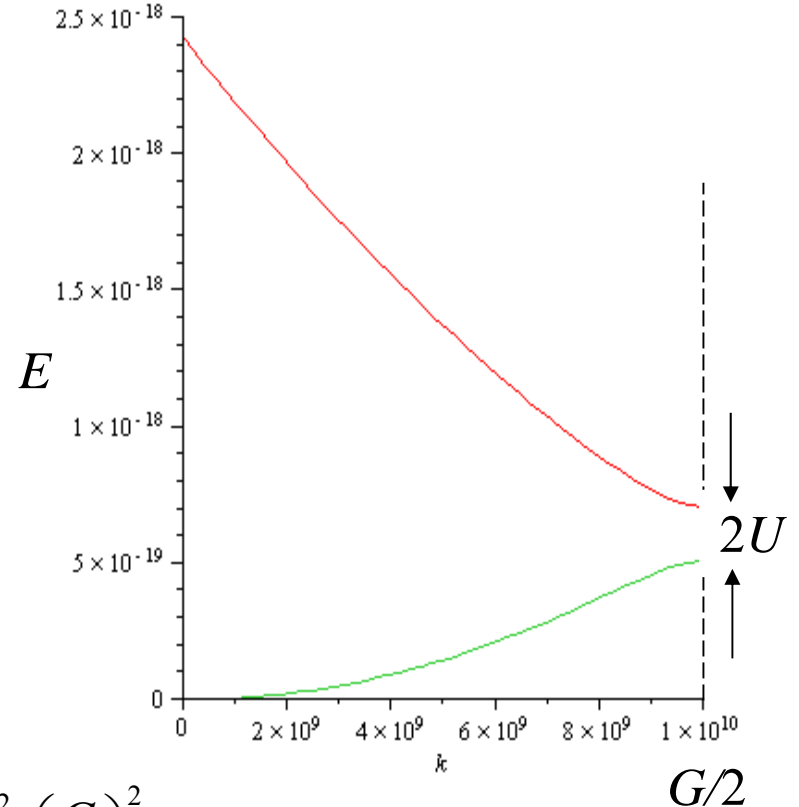
$$\left(\frac{\hbar^2 k^2}{2m} - E \right) C_k + \sum_G U_G C_{k-G} = 0$$

For just 2 terms

$$\begin{bmatrix} \frac{\hbar^2 k^2}{2m} - E & U \\ U & \frac{\hbar^2 (k-G)^2}{2m} - E \end{bmatrix} \begin{bmatrix} C_k \\ C_{k+G} \end{bmatrix} = 0$$

Near the Brillouin zone boundary $k \sim G/2$

$$\begin{bmatrix} \frac{\hbar^2 \left(\frac{G}{2}\right)^2}{2m} - E & U \\ U & \frac{\hbar^2 \left(\frac{G}{2}\right)^2}{2m} - E \end{bmatrix} \begin{bmatrix} C_k \\ C_{k+G} \end{bmatrix} = 0$$



$$E = \frac{\hbar^2 \left(\frac{G}{2}\right)^2}{2m} \pm U$$

Review: Molecules

Start with the full Hamiltonian

$$H = -\sum_i \frac{\hbar^2}{2m_e} \nabla_i^2 - \sum_A \frac{\hbar^2}{2m_A} \nabla_A^2 - \sum_{i,A} \frac{Z_A e^2}{4\pi\epsilon_0 r_{iA}} + \sum_{i<j} \frac{e^2}{4\pi\epsilon_0 r_{ij}} + \sum_{A<B} \frac{Z_A Z_B e^2}{4\pi\epsilon_0 r_{AB}}$$

Use the Born-Oppenheimer approximation

$$H_{elec} = -\sum_i \frac{\hbar^2}{2m_e} \nabla_i^2 - \sum_{i,A} \frac{Z_A e^2}{4\pi\epsilon_0 r_{iA}} + \sum_{i<j} \frac{e^2}{4\pi\epsilon_0 r_{ij}} + \sum_{A<B} \frac{Z_A Z_B e^2}{4\pi\epsilon_0 r_{AB}}$$

Neglect the electron-electron interactions. H_{elec} is then a sum of H_{MO} .

$$H_{MO} = \frac{-\hbar^2}{2m_e} \nabla_1^2 - \sum_A \frac{Z_A e^2}{4\pi\epsilon_0 |r_1 - r_A|}$$

The molecular orbital Hamiltonian can be solved numerically or by the Linear Combinations of Atomic Orbitals (LCAO)


Linear combination of atomic orbitals

Guess that the solution to H_{mo} can be written as a linear combination of atomic orbitals.

$$\Psi_{mo} = c_1 \phi_{1s,A}^H + c_2 \phi_{1s,B}^H + c_3 \phi_{2s,A}^H + c_4 \phi_{2s,B}^H + \dots$$

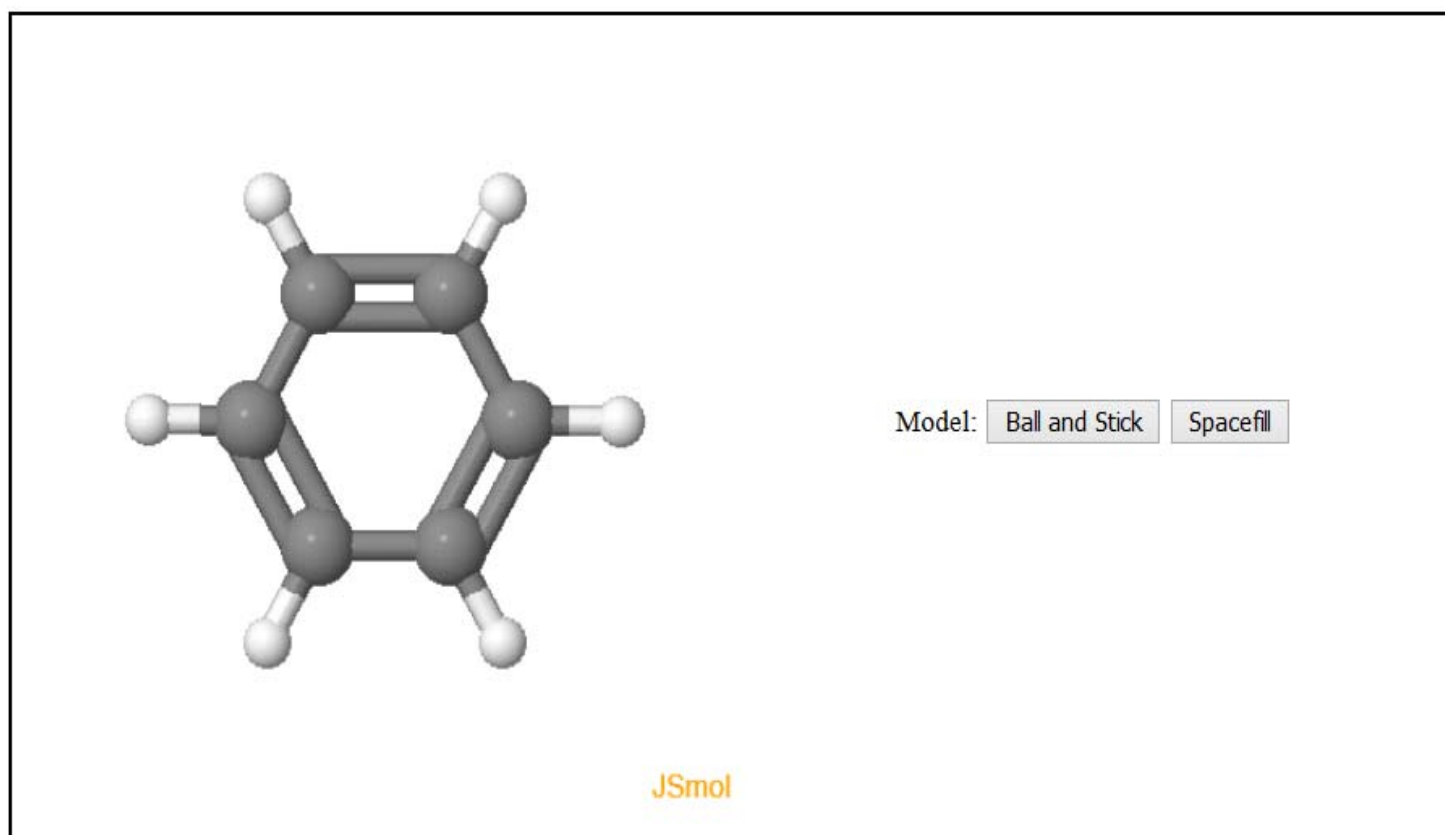
Construct the Hamiltonian matrix.

$$\begin{bmatrix} \langle \phi_{1s,A}^H | H_{mo} | \phi_{1s,A}^H \rangle & \langle \phi_{1s,A}^H | H_{mo} | \phi_{1s,B}^H \rangle \\ \langle \phi_{1s,B}^H | H_{mo} | \phi_{1s,A}^H \rangle & \langle \phi_{1s,B}^H | H_{mo} | \phi_{1s,B}^H \rangle \end{bmatrix} \begin{bmatrix} c_1 \\ c_2 \end{bmatrix} = E \begin{bmatrix} \langle \phi_{1s,A}^H | \phi_{1s,A}^H \rangle & \langle \phi_{1s,A}^H | \phi_{1s,B}^H \rangle \\ \langle \phi_{1s,B}^H | \phi_{1s,A}^H \rangle & \langle \phi_{1s,B}^H | \phi_{1s,B}^H \rangle \end{bmatrix} \begin{bmatrix} c_1 \\ c_2 \end{bmatrix}$$

$$S \approx \begin{bmatrix} 1 & 0 \\ 0 & 1 \end{bmatrix}$$


Molecular orbitals of benzene

Benzene (C_6H_6) consists of 6 carbon atoms in a ring. A hydrogen atom is attached to each carbon atom. The carbon-carbon bond length is 1.40 Å and the carbon-hydrogen bond length is 1.10 Å.



$$\psi_{\text{mo},j} = \frac{1}{\sqrt{N}} \sum_{n=1}^N \exp\left(\frac{i2\pi nj}{N}\right) \phi_{2pz}^C(\vec{r} - \vec{r}_n) \quad j = 1, 2, \dots, N.$$

Tight binding: 1-D chain

$$\psi_k(x) = \frac{1}{\sqrt{N}} \sum_n e^{inka} \phi(x - na)$$

Substitute the tight-binding wave function into the time independent Schrödinger equation.

$$H_{MO}\psi_k = E_k\psi_k$$

Multiply from the left by the atomic orbitals.

$$\langle \phi(x) | \hat{H} | \psi_k(x) \rangle = E \langle \phi(x) | \psi_k(x) \rangle$$

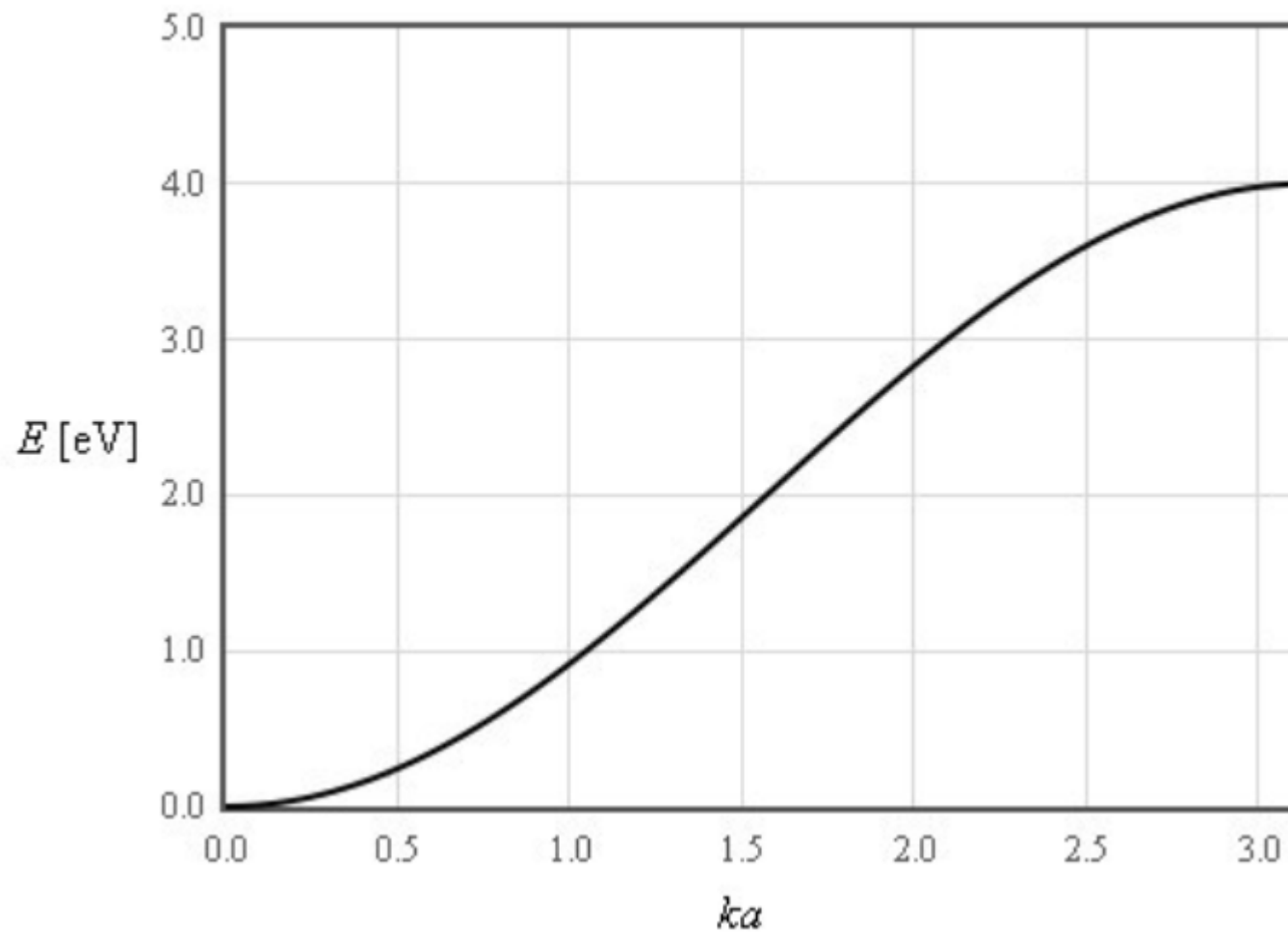
$$\begin{aligned} \langle \phi(x) | \hat{H} | \phi(x - a) \rangle e^{-ika} + \langle \phi(x) | \hat{H} | \phi(x) \rangle + \langle \phi(x) | \hat{H} | \phi(x + a) \rangle e^{ika} + \text{small terms} \\ = E + \text{small terms.} \end{aligned}$$

$$\epsilon = \langle \phi(x) | \hat{H} | \phi(x) \rangle \text{ and } t = -\langle \phi(x) | \hat{H} | \phi(x - a) \rangle$$

$$E = \epsilon - t (e^{-ika} + e^{ika}) = \epsilon - 2t \cos(ka).$$

Tight binding: 1-D chain

$$E = \epsilon - 2t \cos(ka).$$



Tight binding

Tight binding does not include electron-electron interactions

$$H_{MO} = \frac{-\hbar^2}{2m_e} \nabla^2 - \sum_A \frac{Z_A e^2}{4\pi\epsilon_0 |\vec{r} - \vec{r}_A|}$$

$$\psi_k = \frac{1}{\sqrt{N}} \sum_{l,m,n} \exp\left(i\left(l\vec{k} \cdot \vec{a}_1 + m\vec{k} \cdot \vec{a}_2 + n\vec{k} \cdot \vec{a}_3\right)\right) \psi_{\text{unit_cell}}\left(\vec{r} - l\vec{a}_1 - m\vec{a}_2 - n\vec{a}_3\right)$$

$$\psi_{\text{unit_cell}}(\vec{r}) = \sum_i c_i \phi_i(\vec{r} - \vec{r}_i)$$

Atomic wave functions

This is the tight-binding wave function.

$$T_{pqs} \psi_k = \exp\left(i\left(p\vec{k} \cdot \vec{a}_1 + q\vec{k} \cdot \vec{a}_2 + s\vec{k} \cdot \vec{a}_3\right)\right) \psi_k$$

Tight binding, one atomic orbital

$$c_a \langle \phi_a | H_{MO} | \phi_a \rangle + \sum_{\text{nearest neighbors } m} c_m \langle \phi_a | H_{MO} | \phi_m \rangle \exp(i(h\vec{k} \cdot \vec{a}_1 + j\vec{k} \cdot \vec{a}_2 + l\vec{k} \cdot \vec{a}_3)) + \text{small terms}$$
$$= E_k c_a \langle \phi_a | \phi_a \rangle + \text{small terms}$$

For only one atomic orbital in the sum over valence orbitals

$$E_k c_a \langle \phi_a | \phi_a \rangle = c_a \langle \phi_a | H_{MO} | \phi_a \rangle + \sum_{\text{nearest neighbors } m} c_a \langle \phi_a | H_{MO} | \phi_m \rangle \exp(i(h\vec{k} \cdot \vec{a}_1 + j\vec{k} \cdot \vec{a}_2 + l\vec{k} \cdot \vec{a}_3))$$

$$E_k = \varepsilon - t \sum_m e^{i\vec{k} \cdot \vec{\rho}_m}$$

$$\varepsilon = \langle \phi_a(\vec{r}) | H_{MO} | \phi_a(\vec{r}) \rangle$$

On-site energy

$$t = -\langle \phi_a(\vec{r}) | H_{MO} | \phi_a(\vec{r} - \vec{\rho}_m) \rangle$$

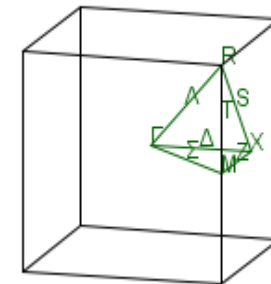
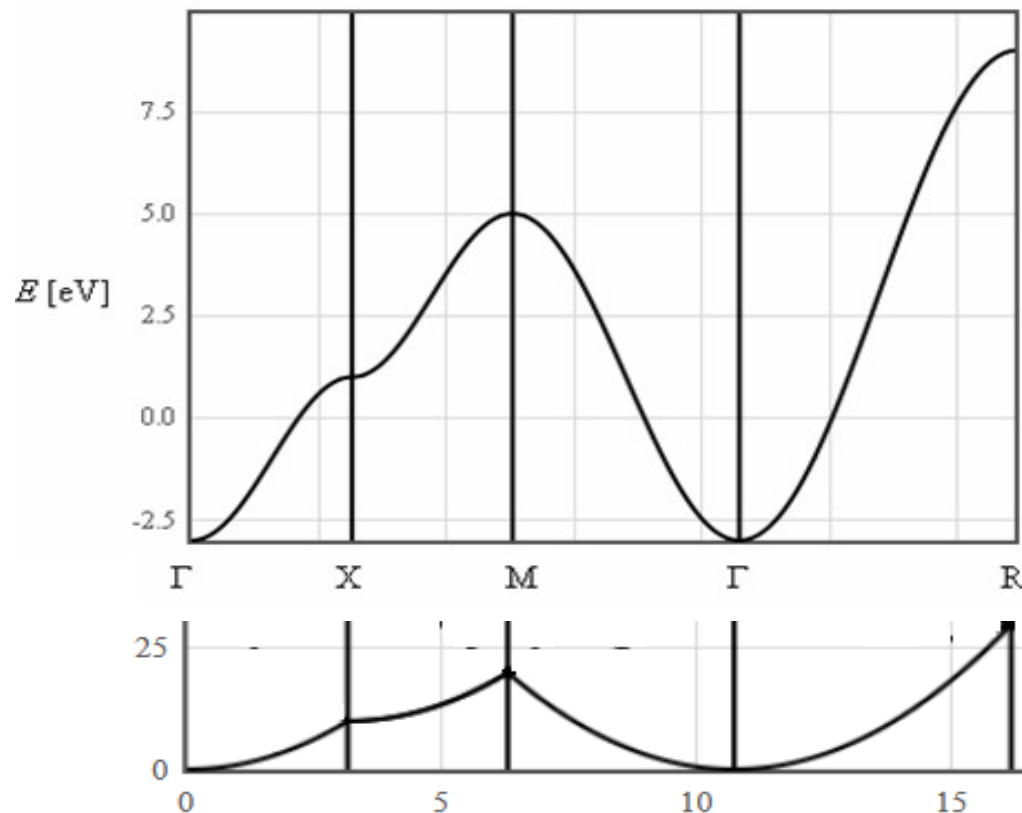
Overlap integral

Tight binding, simple cubic

$$E = \varepsilon - t \sum_{lmn} e^{i\vec{k} \cdot \vec{\rho}_{lmn}}$$

$$E = \varepsilon - t \left(e^{ik_x a} + e^{-ik_x a} + e^{ik_y a} + e^{-ik_y a} + e^{ik_z a} + e^{-ik_z a} \right)$$

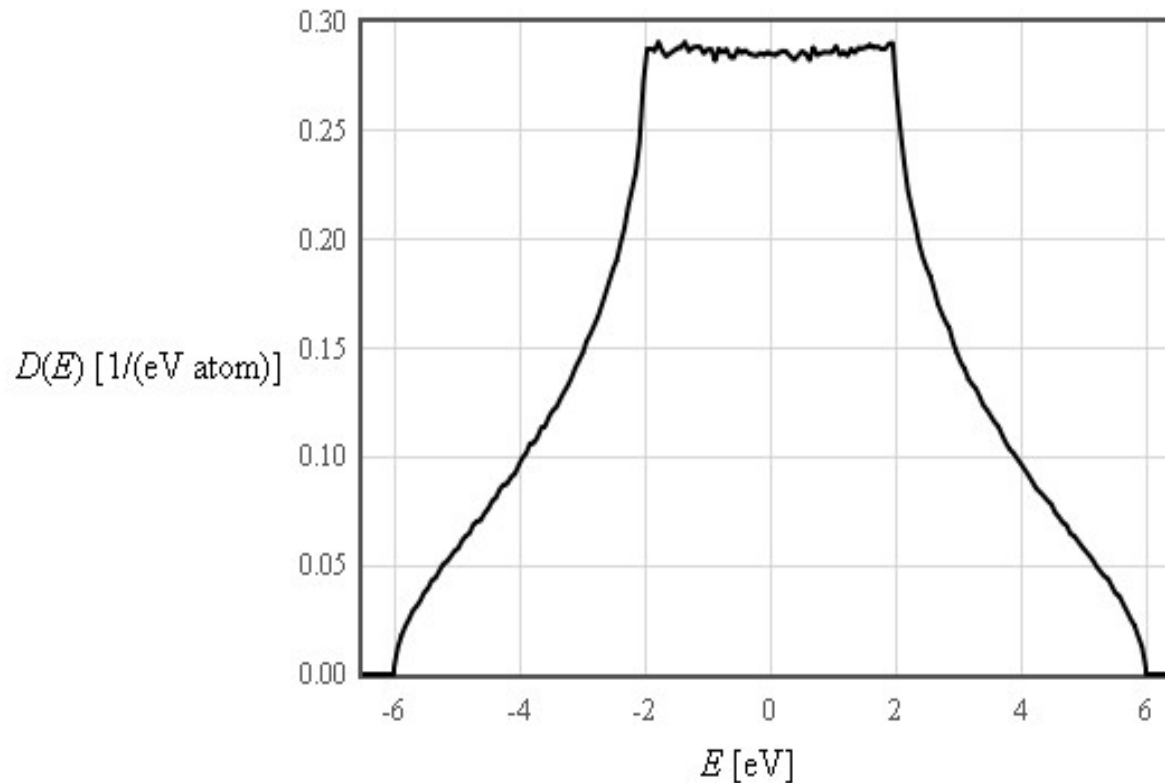
$$= \varepsilon - 2t \left(\cos(k_x a) + \cos(k_y a) + \cos(k_z a) \right)$$



Effective mass $m^* = \frac{\hbar^2}{\frac{d^2 E}{dk^2}} = \frac{\hbar^2}{2ta^2}$

Narrow bands \rightarrow high effective mass

Density of states (simple cubic)



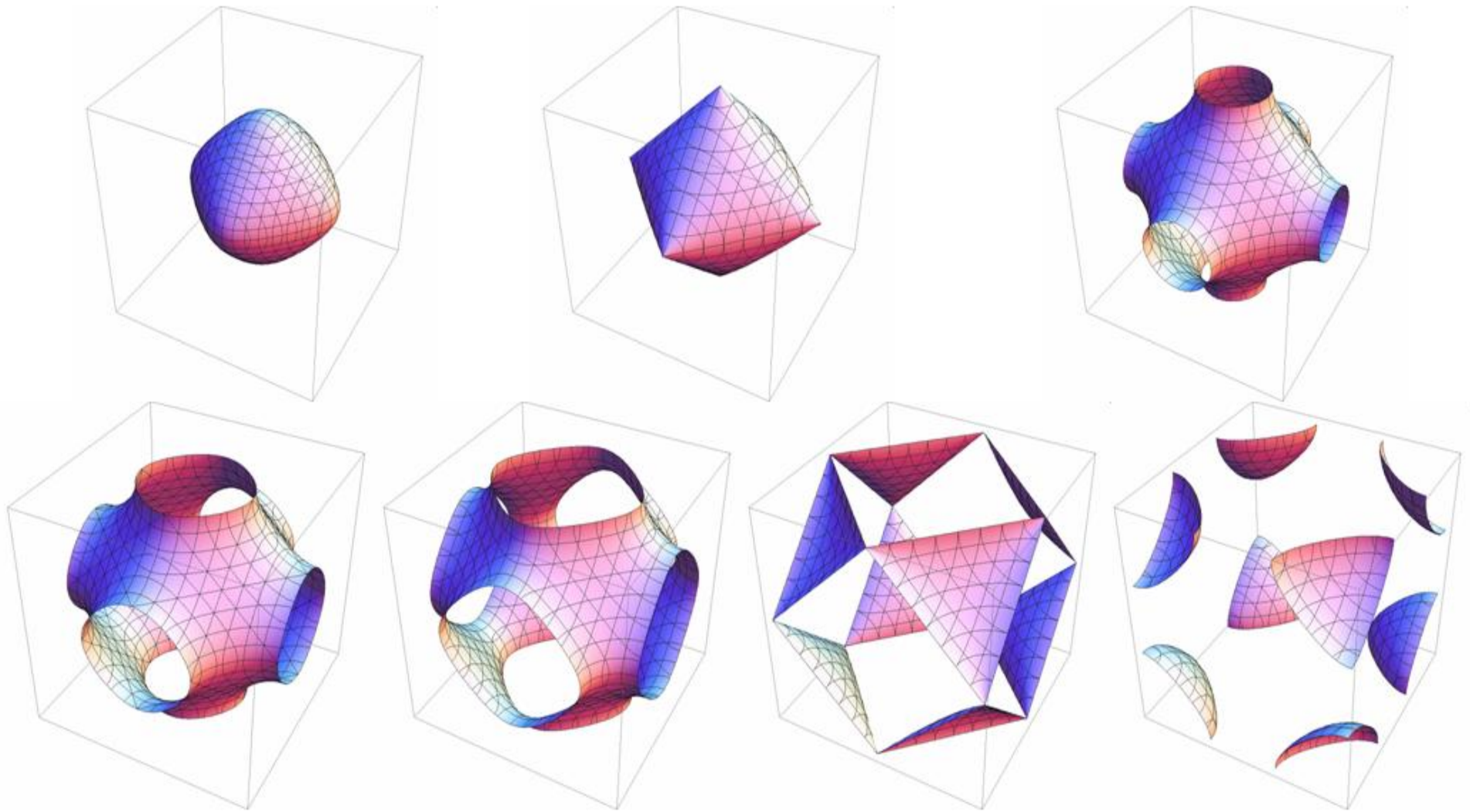
Calculate the energy for every allowed k in the Brillouin zone

$$E = \varepsilon - 2t \left(\cos(k_x a) + \cos(k_y a) + \cos(k_z a) \right)$$

<http://lamp.tu-graz.ac.at/~hadley/ss1/bands/tbtable/tbtable.html>

Tight binding, simple cubic

$$E = \varepsilon - 2t(\cos(k_x a) + \cos(k_y a) + \cos(k_z a))$$



Christian Gruber, 2008