

Institute of Solid State Physics

Technische Universität Graz

Etching

Franssila: Chapters 11, 16, 20, & 21

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SABRE[®] PRODUCT FAMILY

Technology: Electrochemical Deposition (ECD) Solutions: Interconnect, Advanced Memory

http://www.lamresearch.com/products/deposition-products





Lam's market-leading ALTUS systems combine CVD and ALD technologies to deposit the highly conformal films needed for advanced tungsten metallization applications. Nucleation layer formed using Lam's Pulsed Nucleation Layer (PNL) ALD process and in-situ bulk CVD fill.

http://www.lamresearch.com/products/deposition-products

Etching

Wet chemical etching Ion milling Reactive ion etching Chemical-Mechanical Polishing

Wet etching





Etchant

Etch stop (DI water)

etching rate, anisotropy, selectivity

Wet Etching



Figure 11.8 Wet etching tank. Courtesy VTT

Spray etching

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http://www.slideshare.net/gkdelhi8/slide-25-36278815

http://www.cleanroom.byu.edu/wet_etch.phtml

>

Wet Chemical Etching of Metals and Semiconductors

Etch rate depends on deposited m

A comprehensive list of etchants for over 50 different me specified.

Aluminum

s

Aluminum Gallium Arsenide

Aluminum Trioxide / Alumina / Sapphire

Antimony

Bismuth

Brass

Bronze

Carbon

Chromium

Cobalt

Copper

Epoxies

Gallium Arsenide

Germanium

Gold

Hafnium

Indium

Indium Gallium Arsenide

Indium Gallium Phosphide

Indium Phosphide

Indium Phoenhide Ovide Etchante

| Concentrations | Etchants | Rate (angstroms/sec) | Temperature/Other |
|------------------------|--|----------------------|----------------------|
| 1:1 | H ₂ O : HF | | |
| 1:1:1 | HCI: HNO3: H2O | | |
| dilute or concentrated | HCI | | |
| | H ₃ PO ₄ : HNO ₃ : HAc | | |
| 19:1:1:2 | H_3PO_4 : HAc : HNO ₃ : H_2O | 40 | |
| 3:1:3:1 | H_3PO_4 : HAc : HNO ₃ : H_2O | 8.7 @ >RT | @ 40 C <4 min/micror |
| 4:4:1:1 | H_3PO_4 : HAc : HNO_3 : H_2O | 5.6 | |
| 15:0:1:1-4 | H_3PO_4 : HAc : HNO_3 : H_2O | 1500 | 40 C |
| 8:1:1 | H ₃ PO ₄ : H ₂ O ₂ : H ₂ O | 100 | @ 35C |
| 3:1:5 | H ₃ PO ₄ : H ₂ O : glycerin | | |
| 69 : 131 | HCIO ₄ : HAc | | |
| 4:1:5 | HCI : FeCle : H ₂ O | | |
| | FeCl ₃ : H ₂ O | | 100 F |
| 10% | K ₃ Fe(CN) ₈ | 100 | |
| | KOH : K ₃ Fe(CN) ₈ : K ₂ B ₄ O ₇ .4H ₂ O | | |
| 2:3:12 | KMnO ₄ : NaOH : H ₂ O | | |
| 1:1:3 | $NH_4OH : H_2O_2 : H_2O$ | | |
| 20% | NH-SO. | | |

Acid safety

Acid Safety

The following is the manual used to train cleanroom personnel about handling and storing acids:

Operating Instructions



Read the MSDS Know which precautions to take Dispose of acids properly

http://www.cleanroom.byu.edu/acid_safety.phtml

Solvent safety

Solvents used in the IML:

| Chemical | Abbreviation | Fire Hazard | Toxicity Hazard | TLV ppm | Odor Threshold | Toxic Effects |
|---|--------------------------|----------------|--------------------|------------|--------------------------------------|--|
| Acetone | ACE | Extreme | Low | 750 | 140 ppm (sweet/fruity) | Irritates eyes, nose and throat; headaches; skin dryness |
| Freon | TF | Low | Low | 1000 | Variable | Dries skin; light headedness |
| Isopropyl Alcohol | IPA | Extreme | Low | 400 | 20 ppm (sharp/musty) | Dries skin; irritates eyes, nose and throat; drowsiness |
| Methyl Isoamylketone | MIAK | Moderate | Extreme | 50 | 0.05 ppm (sweet/sharp) | Irritates eyes, nose and throat; may cause weakness, dizziness, lightheadedness, nausea, vomiting, or kidney damage |
| Methyl Isobutylketone | МІВК | Extreme | Extreme | 50 | 0.3 ppm (sweet/sharp) | Irritates eyes, nose and throat; may cause weakness, dizziness, lightheadedness, nausea, vomiting, or kidney damage |
| Methyl Ethylketone | MEK | Extreme | Extreme | 200 | 2-100 ppm (misty) | Irritation of eyes and nose; intoxication, headache, and dizziness |
| Ethyl Lactate | Positive Photo Resist | Moderate | Low | None | None (fruity/ester) | Combustible liquid; skin, eye, respiratory irritant; nervous system toxin |
| Propylene Glycol Monomethyl Ether Acetate | PGMEA | Moderate | Low | None | Very low (slightly sweet odor) | Irritant; may cause itching, redness and burns to skin; ingestion may cause diarrhea, kidney and liver damage |

Isotropic and anisotropic etching



Isotropic



Anisotropic

http://mmadou.eng.uci.edu/Book/Q_Chap4.htm

KOH etching of silicon





Figure 20.1 Anisotropic wet-etched profiles in <100> wafer. The sloped sidewalls are the slow etching (111) planes; the horizontal planes are (100). Etching will terminate if the slow etching (111) planes meet

Self limiting depth:
$$d = \frac{W}{\sqrt{2}}$$

KOH etches Si $\{110\} > \{100\} > \{111\}$, producing a characteristic anisotropic V-etch, with sidewalls that form a 54.7° angle with the surface (35.3° from the normal).

http://www.ece.uncc.edu/research/clean_room/fabprocesses/KOH-EtchingAndDecon.pdf





Printer friendly version (here)



Etching through a wafer can take hours

- SAFETY OPERATOR MANUAL (HERE) / PROCÉDURE DE SÉCURITÉ POUR OPÉRER SUR LES WETBENCH DU CMI A LIRE OBLIGATOIREMENT (ICI)
- RESIST IS TOTALLY FORBIDEN INTO THESE BATHES
- THE KOH ETCH RATE COULD VARY: FOR ACCURATE ETCHING PLEASE CALIBRAT IT BEFORE PROCESSING LIVE WAFERS

https://cmi.epfl.ch/etch/PladeKOH.php

KOH etching of silicon

The <111> planes are etched 200 times slower than <100> planes.



Figure 1: Typical profile obtains after Si <100> etching

| ۳°C | Si Etch Rate (µm/h) | Selectivity Si/SiO2 | Bath density |
|-----|---------------------|-----------------------|--------------|
| 60 | 18.7 (±2) | 290 | 1.38 |
| | | Table 1: 40% KOU both | |

Table 1: 40% KOH bath

| ℃ | Si Etch Rate (µm/h) | Selectivity Si/SiO2 | Selectivity Si/Si3N4 | bath density |
|----|---------------------|---------------------|----------------------|--------------|
| 60 | 25 | 458:1 | more than 25000:1 | 1.20 |
| 70 | 42 | 349:1 | more than 25000:1 | 1.20 |
| 80 | 74 | 277:1 | more than 25000:1 | 1.21 |
| 90 | 120 | 204:1 | more than 25000:1 | 1.22 |

Table 2: 23% KOH bath

https://cmi.epfl.ch/etch/PladeKOH.php

Other anisotropic etchants for silicon

| Etchant | Operating temp (°C) | R ₁₀₀ (µm/min) | S=R ₁₀₀ /R ₁₁₁ | Mask materials |
|--|---------------------|---------------------------|--------------------------------------|--|
| Ethylenediamine pyrocatechol (EDP) ^[2] | 110 | 0.47 | 17 | SiO ₂ , Si ₃ N ₄ , Au, Cr, Ag, Cu |
| Potassium hydroxide/lsopropyl alcohol (KOH/IPA) | 50 | 1.0 | 400 | Si ₃ N ₄ , SiO ₂ (etches at 2.8 nm/min) |
| Tetramethylammonium hydroxide (TMAH) ^[3] | 80 | 0.6 | 37 | Si ₃ N ₄ , SiO ₂ |

EDP (an aqueous solution of ethylene diamine and pyrocatechol), displays a <100>/<111> selectivity of 17X, does not etch silicon dioxide as KOH does, and also displays high selectivity between lightly doped and heavily boron-doped (p-type) silicon.

Tetramethylammonium hydroxide (TMAH) presents a safer alternative than EDP, with a 37X selectivity between {100} and {111} planes in silicon.

http://en.wikipedia.org/wiki/Etching_%28microfabrication%29

HF etching of SiO₂

$$SiO_2 + 6 HF = 2H^+ SiF_6^{2-} + 2H_2O$$

Stops at the silicon surface and leaves the surface hydrogen passivated.

HF is dangerous and you require special training before using it. Larger labs have a dedicated HF station.

HF reacts with glass, concrete, metals, water, oxidizers, reducers, alkalis, combustibles, organics and ceramics. It must be kept in special polyethylene or fluorocarbon plastic containers and special tools are used.

Etch-stop techniques

p+ etch stop

silicon highly doped (> 10^{19} cm⁻³) with boron etches very slowly

Etch stop with buried masking layers

implant O, N, or C to make SiO₂, SiC, or SiN_x

Electrochemically controlled pn etch stop

The voltage drops across the reverse biased junction until the n region is exposed and then the potential drop at the surface oxidizes the silicon and stops the etching.



http://memslibrary.com/guest-articles/47-silicon-etching/4-etch-stop-techniques-for-etching-of-silicon-in-alkaline-solutions.html and the solution of the so

Plasma etching



The plasma activates the etching gas which reacts at the surface to form a gaseous product.

(In PECVD a solid product is formed.)

Etchants for common microfabrication materials

| Material to be etched | Wet etchants | Plasma etchants |
|---|---|--|
| Aluminium (Al) | 80% phosphoric acid (H_3PO_4) + 5% acetic acid + 5% nitric acid (HNO_3) + 10% water (H_2O) at 35–45 °C ^[4] | Cl ₂ , CCl ₄ , SiCl ₄ , BCl ₃ ^[5] |
| Indium tin oxide [ITO] (In ₂ O ₃ :SnO ₂) | Hydrochloric acid (HCl) + nitric acid (HNO ₃) + water (H ₂ O) (1:0.1:1) at 40 $^{\circ}C^{[6]}$ | |
| Chromium (Cr) | "Chrome etch": ceric ammonium nitrate ((NH₄)₂Ce(NO₃)₆) + nitric acid (HNO₃)^[7] Hydrochloric acid (HCl)^[7] | |
| Gallium Arsenide (GaAs) | Citric Acid diluted (C₆H₈O₇ : H₂O, 1 : 1) + Hydrogen Peroxide (H₂O₂)+ Water (H₂O) | • Cl_2 , CCl_4 , $SiCl_4$, BCl_3 , CCl_2F_2 |
| Gold (Au) | Aqua regia, lodine and Potassium lodide solution | |
| Molybdenum (Mo) | | CF ₄ ^[5] |
| Organic residues and photoresist | Piranha etch: sulfuric acid (H_2SO_4) + hydrogen peroxide (H_2O_2) | O ₂ (ashing) |
| Platinum (Pt) | Aqua regia | |
| Silicon (Si) | Nitric acid (HNO₃) + hydrofluoric acid (HF)^[4] Potassium hydroxide (KOH) Ethylenediamine pyrocatechol (EDP) Tetramethylammonium hydroxide (TMAH) | • CF ₄ , SF ₆ , NF ₃ ^[5] • Cl ₂ , CCl ₂ F ₂ ^[5] |
| Silicon dioxide (SiO ₂) | Hydrofluoric acid (HF)^[4] Buffered oxide etch [BOE]: ammonium fluoride (NH₄F) and hydrofluoric acid (HF)^[4] | CF ₄ , SF ₆ , NF ₃ ^[5] |
| Silicon nitride (Si ₃ N ₄) | 85% Phosphoric acid (H₃PO₄) at 180 °C^[4] (Requires SiO₂ etch mask) | CF ₄ , SF ₆ , NF ₃ , ^[5] CHF ₃ |
| Tantalum (Ta) | | CF4 ^[5] |
| Titanium (Ti) | Hydrofluoric acid (HF) ^[4] | BCI3 ^[8] |
| Titanium nitride (TiN) | Nitric acid (HNO₃) + hydrofluoric acid (HF) SC1 Buffered HF (bHF) | |

Plasma etching

The same equipment can be used for

plasma etching plasma cleaning surface modification

Leaves less residue than wet etching. The products are volatile.

Ion Milling

Ions (typically Ar) are accelerated at the substrate. No chemical reaction Selectivity ~ 1:1 High vacuum

Will etch anything



http://hitachi-hta.com/products/electron-ion-and-probe-microscopy/ion-beam-milling/im4000-ion-milling-system

Reactive Ion Etching (RIE)





Combines physical ion milling with chemical etching. Is faster and more selective than ion milling.

Isotropic and Anisotropic Plasma Etching



Figure 11.4 Isotropic (left) and anisotropic etch profiles (right)

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You can use plasma etching to etch isotropically and anistotropically.

Isotropic and Anisotropic Plasma Etching



Figure 21.15 Buried microchannels: (a) anisotropic DRIE, sidewall spacer formation and isotropic DRIE; (b) removal of spacer and conformal CVD. SEM micrograph from de Boer *et al.* (2000) by permission of IEEE

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Channels used for microneedles.

Microbolometer



Figure 11.15 Bolometer fabrication process: left, resistor lithography and etching; right, second lithography, oxide etching and silicon isotropic etching



Figure 11.16 Spiral antenna microbolometer: silicon is isotropically etched to release the narrow resistor. SEM courtesy Leif Grönberg, VTT

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Bosch process

Repeat 2 processes over and over

- 1. Etch Si with SF_6 (nearly isotropic)
- 2. Deposit passivation layer C_4F_8

Directional etching at the bottom breaks through the passivation layer.

Short cycles: smooth walls

Long cycles: fast etching





http://en.wikipedia.org/wiki/Deep_reactive-ion_etching

Through-Silicon Via (TSV)



A vertical electrical connection (via) passing completely through a silicon wafer.

Used in 3D integration.

Chemical Mechanical Polishing (CMP)



Woodpile photonic crystal



http://www.sandia.gov/media/photonic.htm

Damascene process



Inlaying of one metal in another

http://en.wikipedia.org/wiki/Damascening#/media/File:Damascening.jpg

Interconnects

22 nm Process



80 nm minimum pitch

14 nm Process



52 nm (0.65x) minimum pitch

http://www.tomshardware.com/reviews/intel-14nm-broadwell-y-core-m,3904.html

Cleaning

Wafer cleaning is about 30% of all processes.

```
Particles - brushes, water jets, shockwaves
cause: lithography defects, pinholes, shorts
Organics - peroxide, O_2 plasma
hydrophobic (inhibits water cleaning)
residues keep subsequent layers from sticking
Metals - Acids (HCl-H<sub>2</sub>O<sub>2</sub> and H<sub>2</sub>SO<sub>4</sub>-H<sub>2</sub>O<sub>2</sub>)
Oxide - HF
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Clean rooms

Filtered air to remove particles Overpressure maintained to blow dirt out Controlled temperature and humidity Important to make processes reproducible



Figure 35.5 Cleanroom and gray area: ISO 3 (Class 1) area for wafer processing, ISO 6 (Class 1000) turbulent flow in service aisle Reproduced from Whyte (2001) by permission of John Wiley & Sons, Ltd



http://www.cleanroominnovation.com

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Particles

Flakes from chamber walls Wear of mechanical parts



The concentration of particles increases exponentially as their size decreases.

Foup (Front Opening Universal Pod)



- · FIMS Door/Automated operation
- Manual open/close function
- · Robust structure for perfect seal during transportation
- · Design to minimize Cleaning and Drying Cycle Time
- Conforms to SEMI Standard M31

Dimensions : W389 x L340 x H331 (mm) Weight : 7.5kg(16.9lb) Including 25 wafers 4.3kg(9.7lb) without wafers





https://www.shinpoly.co.jp/business/seimitsu/300gt/index.html

Wafer handling



Wafer transport

https://www.youtube.com/watch?v=-KTKg0Y1snQ

Foup cleaning



Cleaning





Wafer cleaning is a critical function that must be repeated many times during semiconductor manufacturing.

KEY APPLICATIONS

- Particle, polymer, and residue removal
- Photoresist removal
- Backside/bevel cleaning and film removal

Villach/Austria is the global centre for the development and production of all single-wafer spin technology products for back- and front- end-of-line (BEOL/FEOL) cleaning, etching and stripping applications.

Photoresist strip





Strip processes remove photoresist material after it has served to "protect" certain areas of the wafer surface from being altered.

Chemistry and fluid mechanics are important.



http://www.lamresearch.com/products/strip-clean-products

Contact angle



Table 12.6Water contact angles for varioussurfaces and treatments

| Ammonia/peroxide cleaned silicon | 5° |
|----------------------------------|------------------------|
| Oxygen plasma treated SU-8 | $5^{\circ}-40^{\circ}$ |
| Sulfuric acid cleaned silicon | 10° |
| RCA-1 + RCA-2 cleaned silicon | 10° |
| KOH etched silicon | 25° |
| Thermal oxide | 45° |
| Native oxide | 45° |
| Oxygen plasma treated PDMS | 50° |
| HMDS coated silicon | 60° |
| HF dipped silicon | 70° |
| Polyimide | 75° |
| Native SU-8 | 80° |
| Native polystyrene | 90° |
| Native PDMS | 108° |
| ECT (eicosanethiol) | 110° |
| Fluoropolymer | 120° |
| Microstructure + PDMS | 150° |
| Nanostructure + fluoropolymer | 170° |
| | |

Note that all the values are approximate and depend on surface treatment details and duration, and on time delay.

http://en.wikipedia.org/wiki/Contact_angle

Dry cleaning

Etching using plasmas or gases Can be used in a MBE or CVD chamber Ozone/UV for organics HF vapors for oxide Cl_2 for metals Ar milling/sputter cleaning for anything

Drying

DI water -> IPA -> Blow dry N_2

